



KAZAKH NATIONAL UNIVERSITY

Calculation of catalytic nuclear reactions induced by reactor neutrons and its applications

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Introduction

- **I spent 3 month in Hokkaido University in Dr. Go Chiba's laboratory**
- **He is a head of reactor physics laboratory in HU**

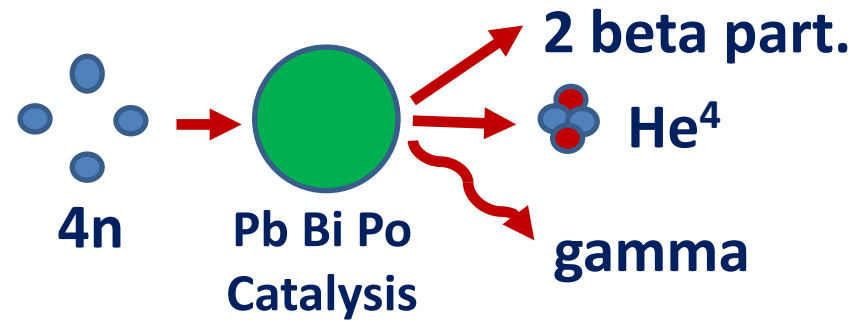
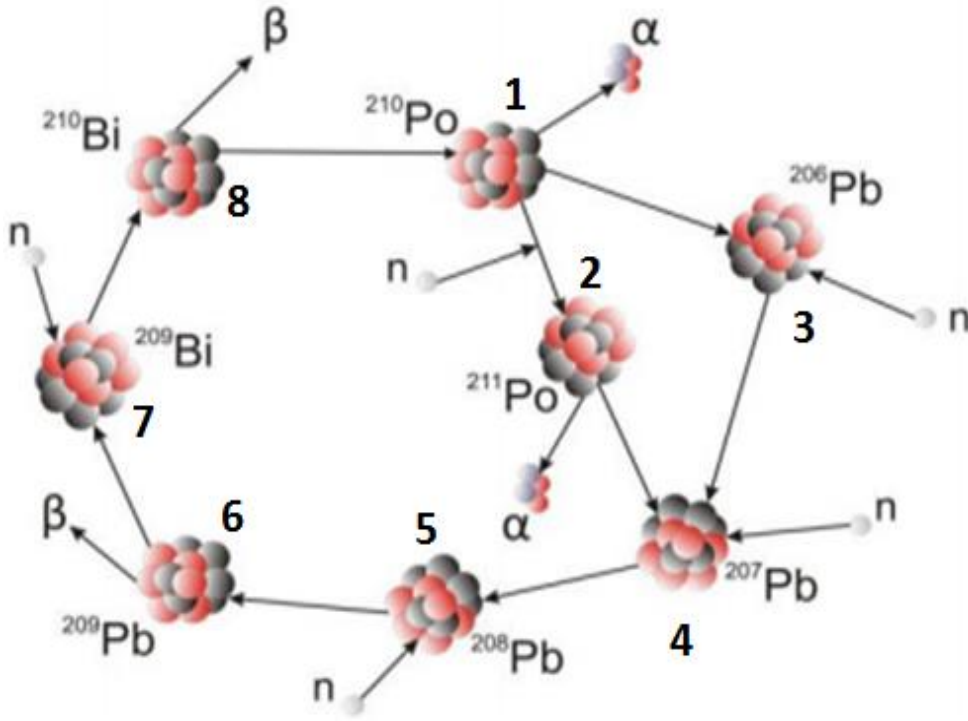


Prof. Go Chiba

Introduction

Neutron catalysis is a chain reaction based on the reaction of a four neutron capture by catalyst nucleus with the restoration of the initial nucleus

Catalytic cycle reaction - four of neutron capture, two beta decay and alpha decay.



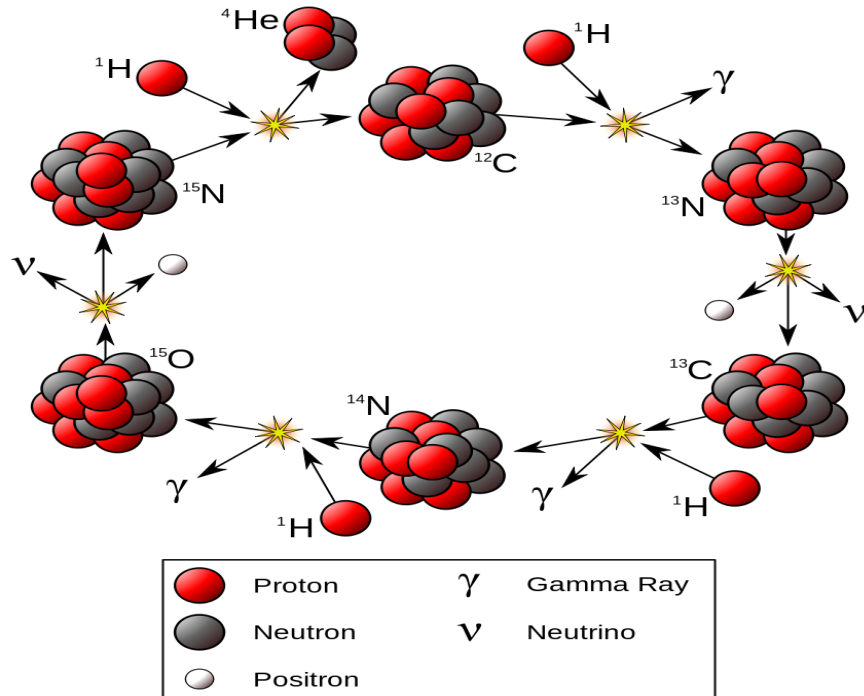
$$\Delta E = (4 * M_n - M_{alfa} = 30 \text{ MeV})$$

Introduction

Analog of neutron catalysis is:

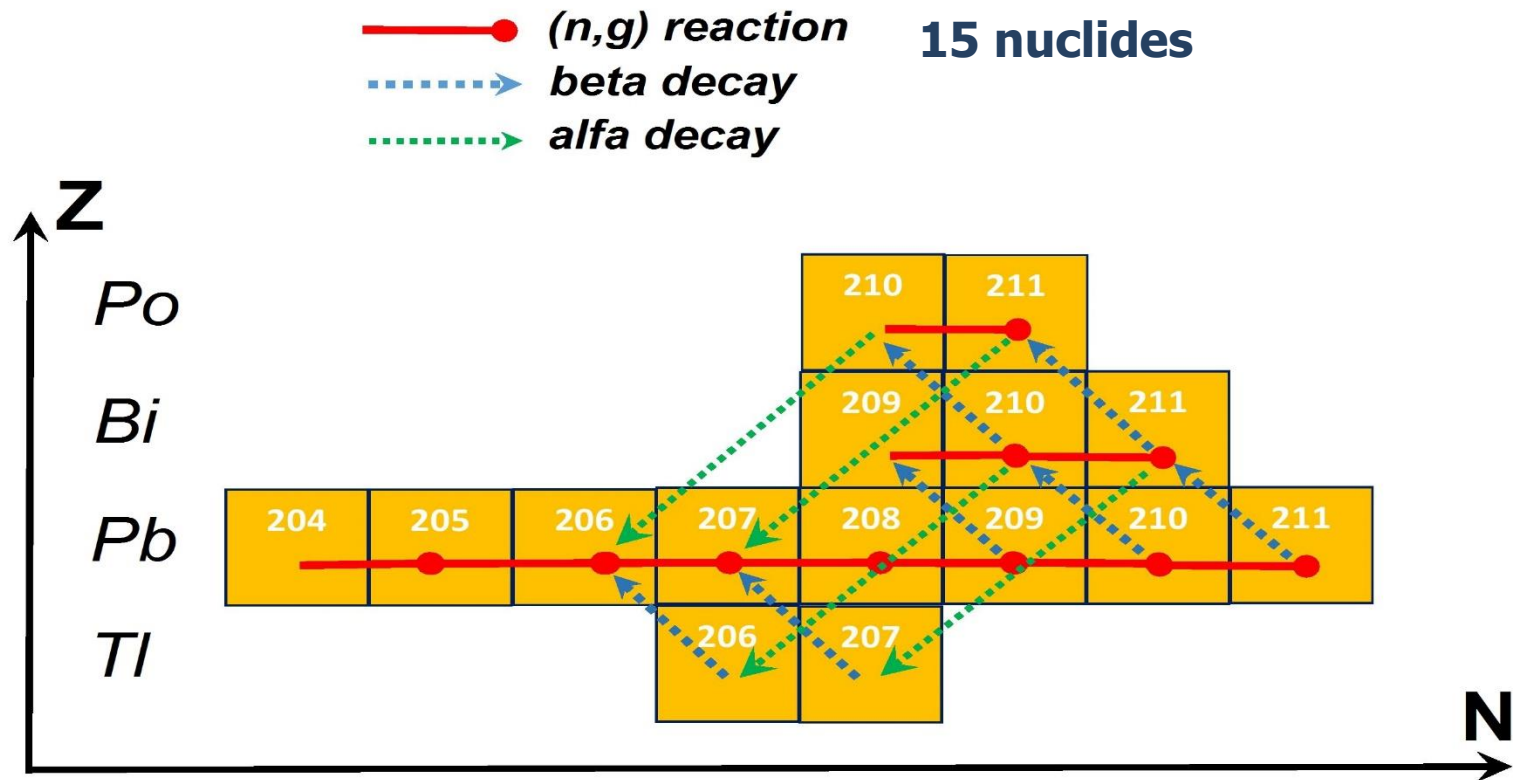
CNO cycle in massive stars

CNO cycle (for **carbon–nitrogen–oxygen**) is fusion reactions by which stars convert hydrogen to helium



Introduction

Neutron catalysis is the catalytic chain reaction with *equilibrium state*.



Calculation

Differential equation for a number density of nuclide i

$$\frac{dN_i}{dt} = \underbrace{-\lambda_i N_i(t) - \sigma_i \phi N_i(t)}_{\text{Decrease of } N_i} + \underbrace{\sum_{j \neq i} \lambda_j P_{j \rightarrow i} N_j(t) + \sum_{j \neq i} \sigma_j \phi Q_{j \rightarrow i} N_j(t)}_{\text{Increase number density of } N_i}$$

$$\frac{d}{dt} \mathbf{N} = \mathbf{A} \mathbf{N}$$

$$\mathbf{N} = (N_1, N_2 \dots N_I)$$

\mathbf{A} – **burn-up** matrix

\mathbf{N} – Nuclide density vector

System of linear equations

$$\mathbf{N}(t) = \mathbf{N}(0) \exp(\mathbf{A}t)$$

The number density in the equilibrium state can be easily obtained if $t \rightarrow \infty$



Calculation

$$\mathbf{N}(t) = \mathbf{N}(0) \exp(\mathbf{A}t)$$

The matrix exponential $\exp(\mathbf{A}t)$ can be numerically calculated with several methods

1. Pade approximation
2. CRAM (Chebyshev Rational Approximation) – developed in VTT technical research center of Finland
3. MMPA – developed in Hokkaido University

$$N_{pq}(x) = \sum_{k=0}^p \frac{(p+q-k)! p!}{(p+q)! k! (p-k)!} x^k$$
$$D_{pq}(x) = \sum_{k=0}^q \frac{(p+q-k)! q!}{(p+q)! k! (q-k)!} (-x)^k$$
$$\exp(\mathbf{A}\Delta t) \approx \frac{N_{pq}(\mathbf{A}\Delta t)}{D_{pq}(\mathbf{A}\Delta t)}$$



One group cross section

In burnup calculation it is convenient to use 1-group energy averaged cross sections

1-group cross section is defined from energy-dependent cross section $\sigma(E)$ and neutron flux $\varphi(E)$.

G is the total number of energy groups

$$\tilde{\sigma} = \frac{\int \sigma(E)\varphi(E)dE}{\int \varphi(E)dE}$$

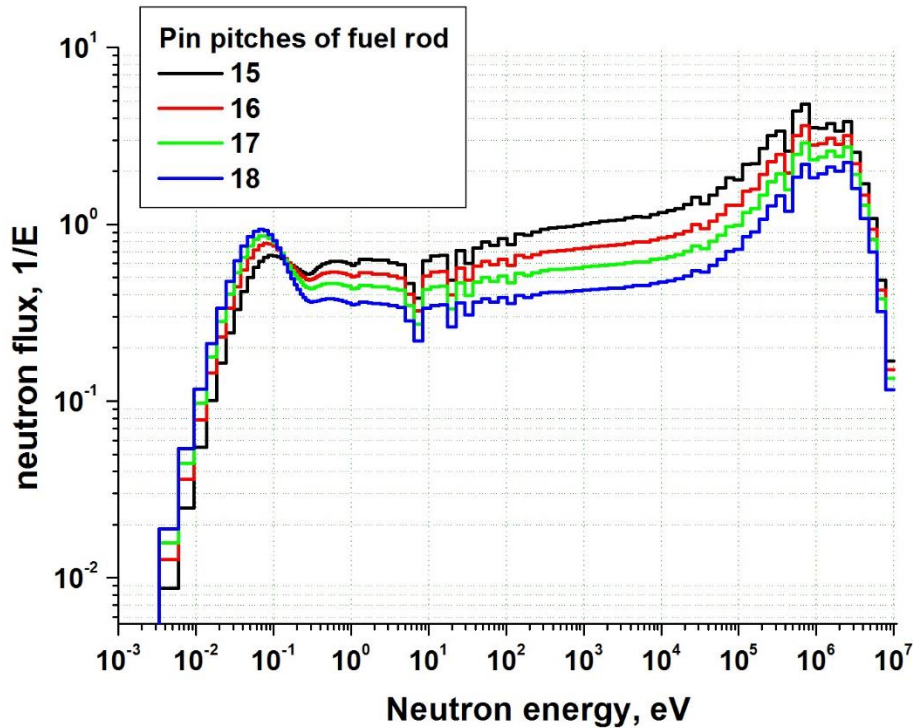
G is the total number of energy groups
 $\varphi(E)$ is energy spectra of neutron flux

$$\tilde{\sigma} = \frac{\sum_{g=1}^G \sigma_g \varphi_g}{\sum_{g=1}^G \varphi_g}$$

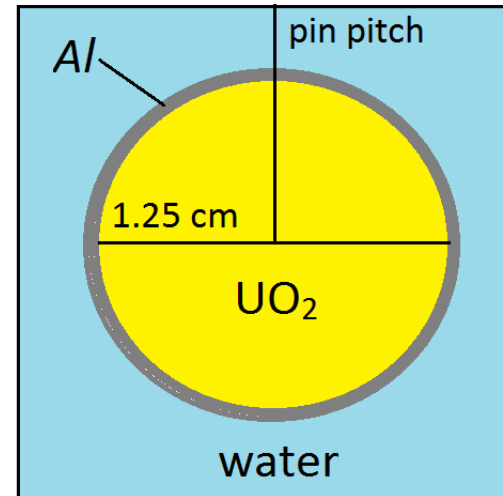


One group cross section

Neutron flux spectra at different pin pitches of the tank-typed critical assembly **TCA** with **CBZ** (**TCA** – is Japanese experimental reactor)



provided by Prof. Go Chiba

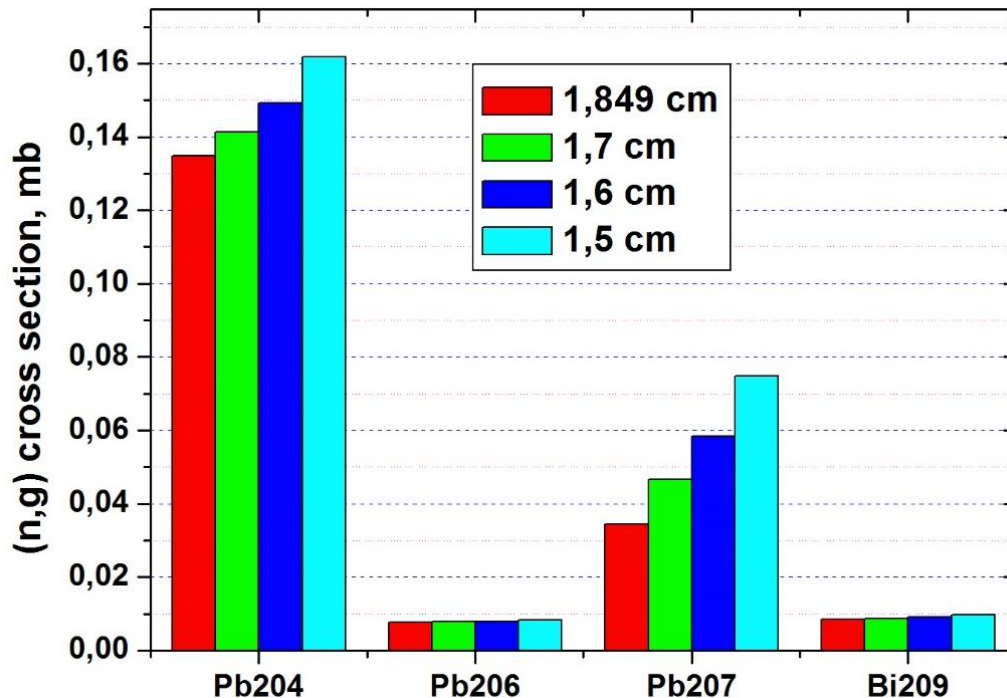


pin cell of TCA

One group cross section

(n,g) cross sections were taken from **JENDL-4.0 library**

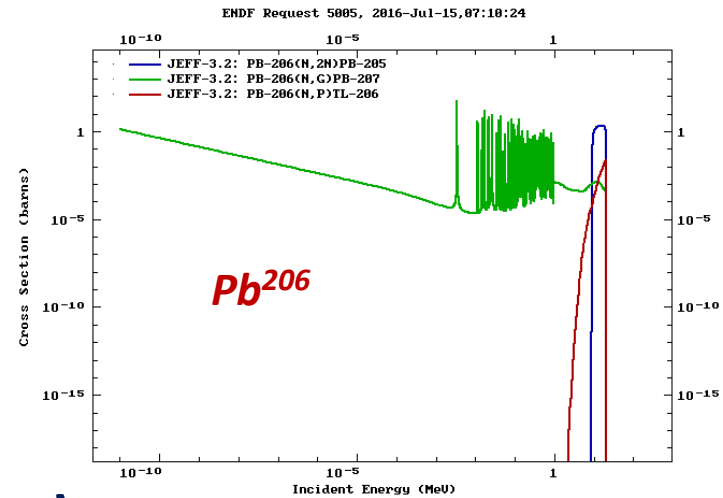
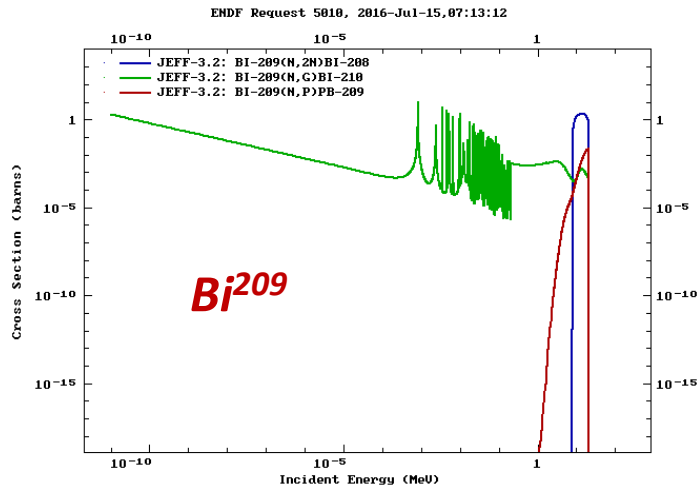
Energy averaged cross sections of stable nuclides



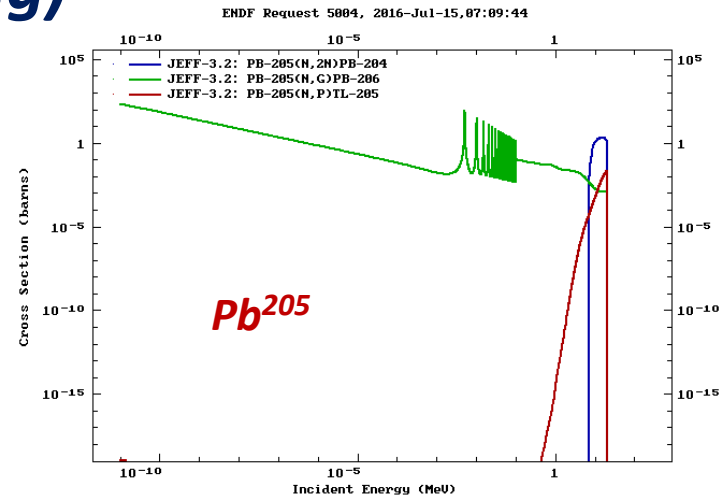
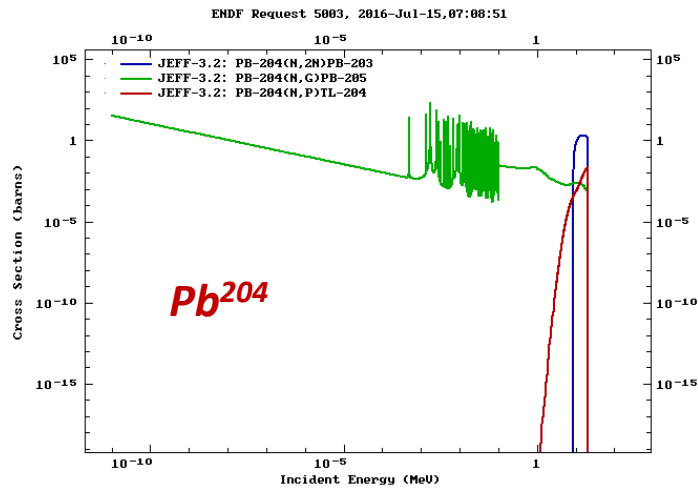
Bi²⁰⁹ Pb²⁰⁴ Pb²⁰⁶ Pb²⁰⁷ Pb²⁰⁹
Tl²⁰⁶ Tl²⁰⁷ Pb²⁰⁵ Pb²¹⁰ Pb²¹¹
Bi²¹⁰ Bi²¹¹ Po²¹⁰ Po²¹¹



One group cross section (<https://www-nds.iaea.org/exfor/endl.htm>)



$\sigma(n,g)$

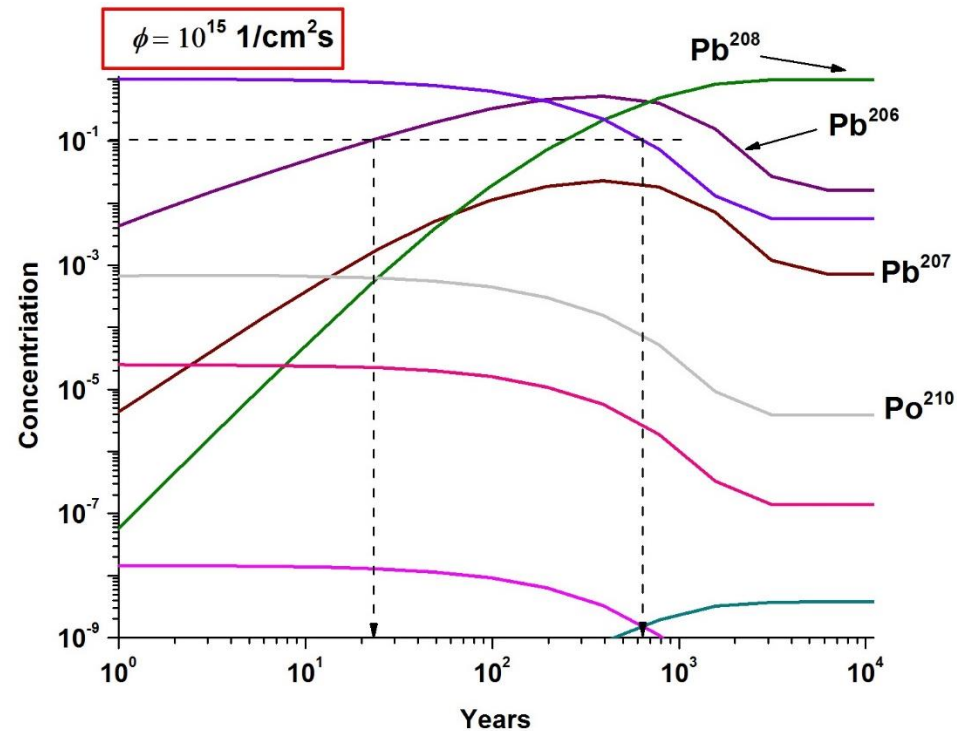
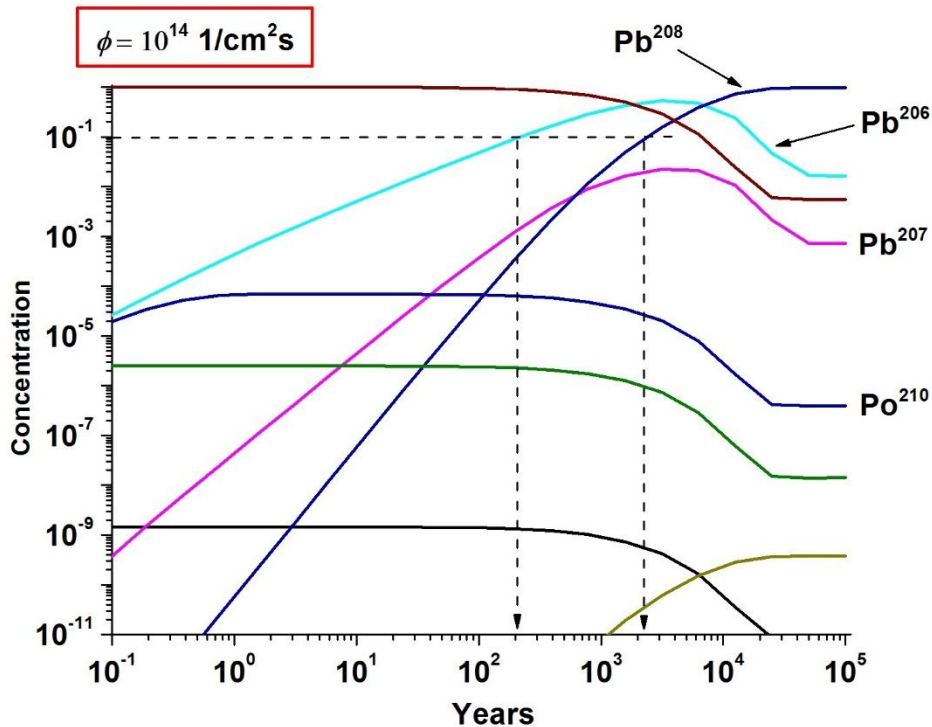


Calculation

Number density change of nuclides under reactor neutron irradiation

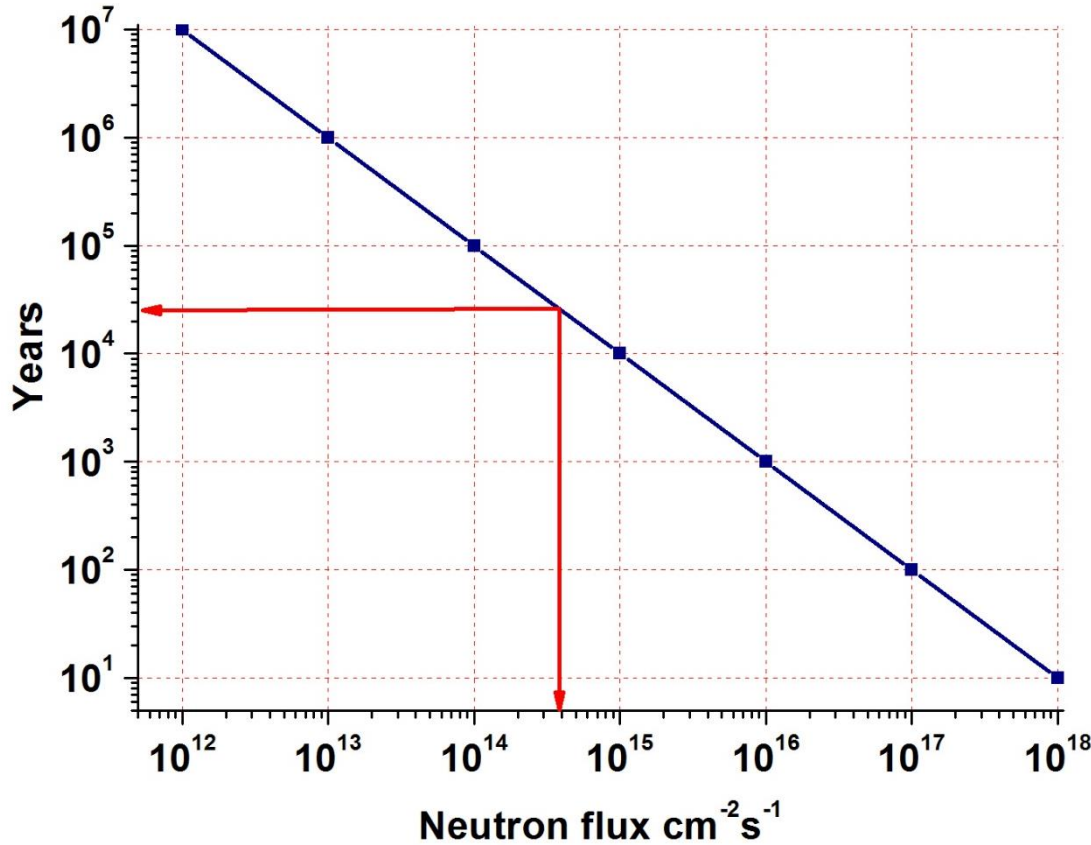
Initial composition is **natural lead**

Pb²⁰⁴	1,4%	Pb²⁰⁷	22,1%
Pb²⁰⁶	24,1%	Pb²⁰⁸	52,4%



Calculation

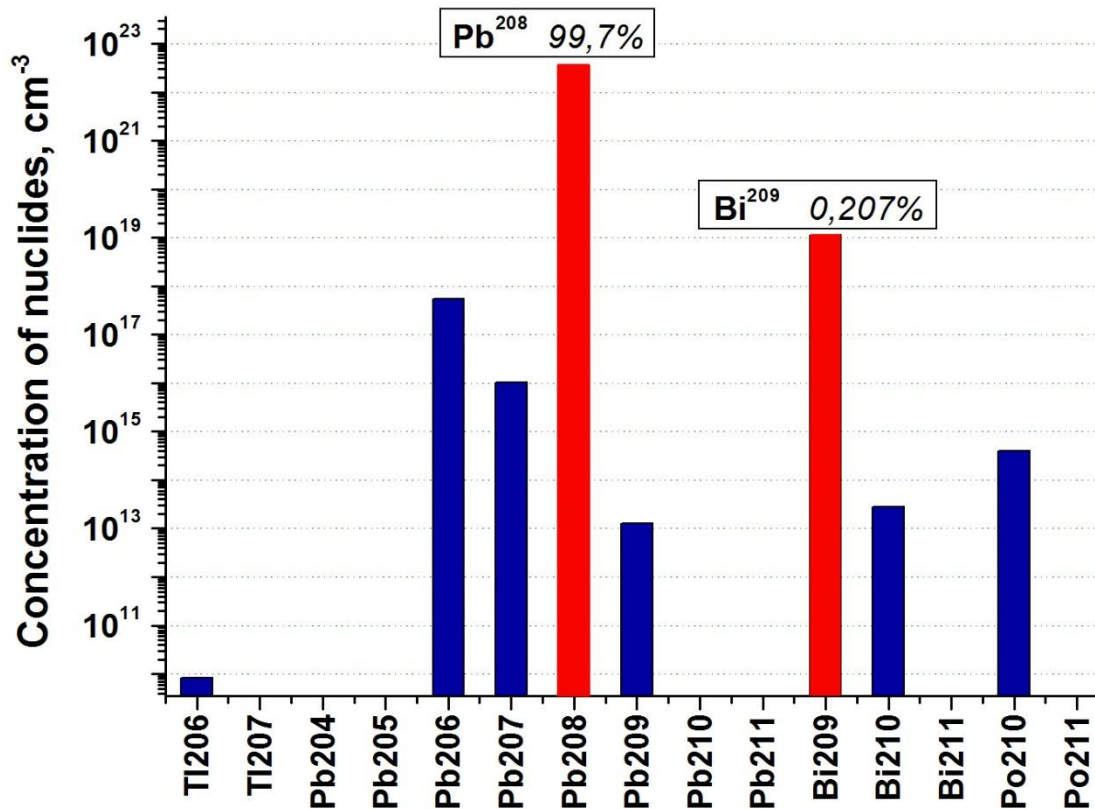
The time needed to reach **the equilibrium state** at different neutron fluxes



Problem is the time period of equilibrium state is too long

Calculation

Number density of nuclides at *equilibrium state*.



The problem is:

It is too difficult to enrich 15 nuclides for proper densities.

The *natural lead* composition is used for calculation because its isotope composition is closer to catalytic material composition

Application

How can catalytic material be used as reactor component?



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Main advantage of catalytic composition is:

- **It is stable under long time neutron irradiation**
- **It can produce additional energy in reactor core**



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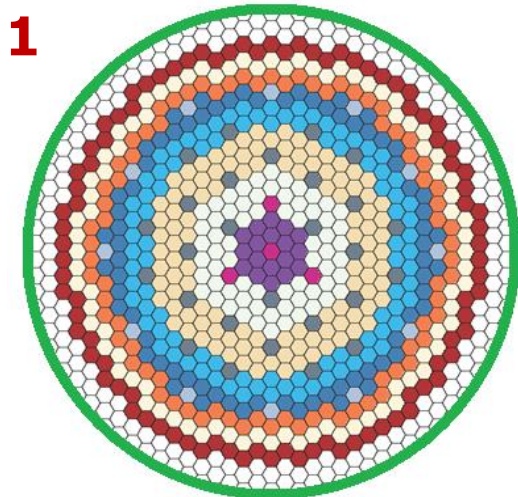


Application

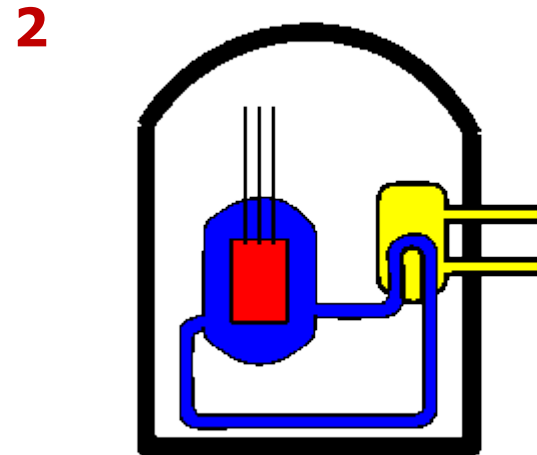
How can catalytic material be used as a reactor component?

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As neutron reflector
except Be



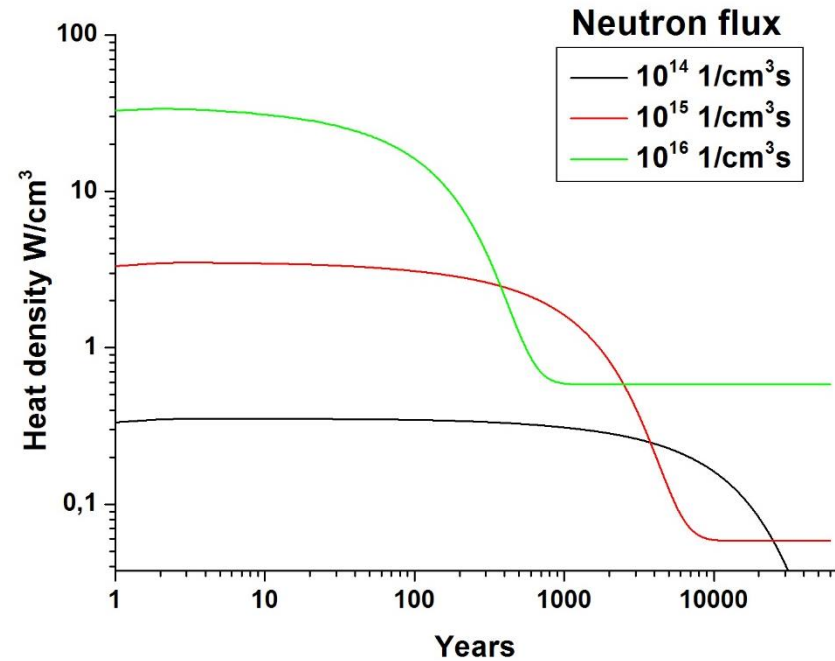
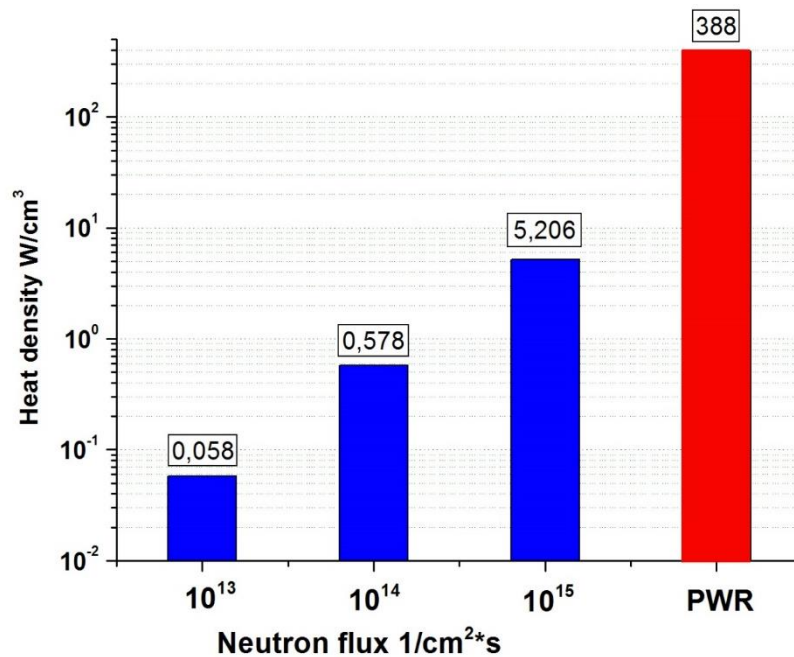
As liquid lead coolant in fast reactor

Heat density results

Total heat generated by decays

E_i is the emitted energy by a decay of nuclide i

$$H(t) = \sum_i E_i \lambda_i N_i(t) + \sum_i E_i' \phi \sigma_i N_i(t)$$



Sensitivity analysis of nuclear data



Sensitivity analysis of nuclear data

$$S^i = \frac{dN_i(T)}{N_i(T)} \bigg/ \frac{d\sigma}{\sigma} = \frac{\sigma}{N_i(T)} \cdot \frac{dN_i(T)}{d\sigma}$$

\mathbf{M} is burnup matrix

\mathbf{N}^* is adjoint number density vector

T is the time after irradiation

σ capture cross section (n,g)

$$\frac{dN_i(T)}{d\sigma} = \int_0^T \left(\mathbf{N}^* \frac{d\mathbf{M}}{d\lambda} \mathbf{N} \right) dt$$

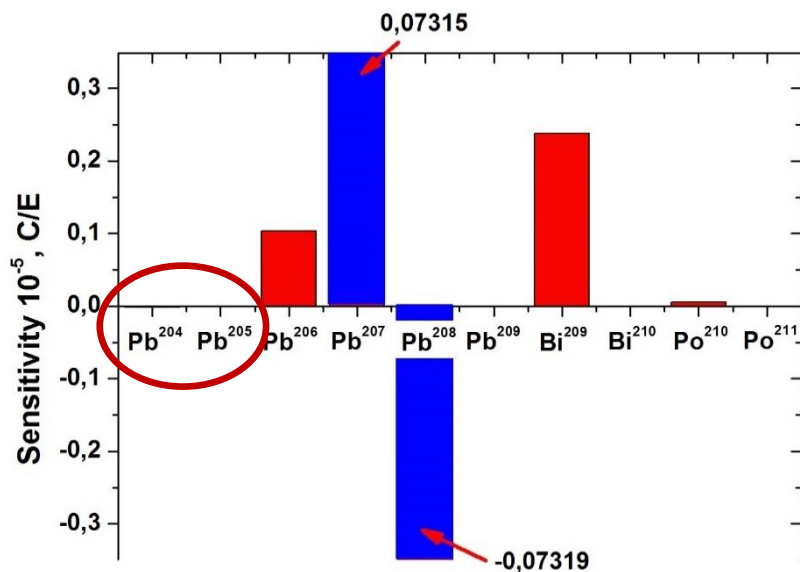
Derivative term can be derived using *General Perturbation Theory*. The calculation method was developed by Prof. Go Chiba in JAEA. (Sensitivity Analysis of Fission Product Concentrations for LWR Burned Fuel)

Note: The position dependence of concentrations $N(r)$ and the energy dependence of cross sections $\sigma(E)$ are ignored here.

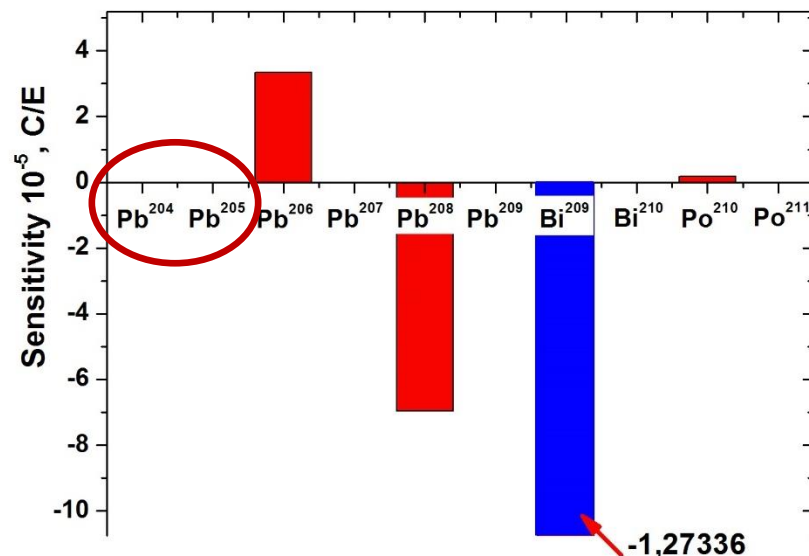


Sensitivity analysis of nuclear data

Sensitivity analysis of nuclear data can show which nuclear data is important to results.



Sensitivities of the $Pb-208$ concentration to the (n,g) cross sections

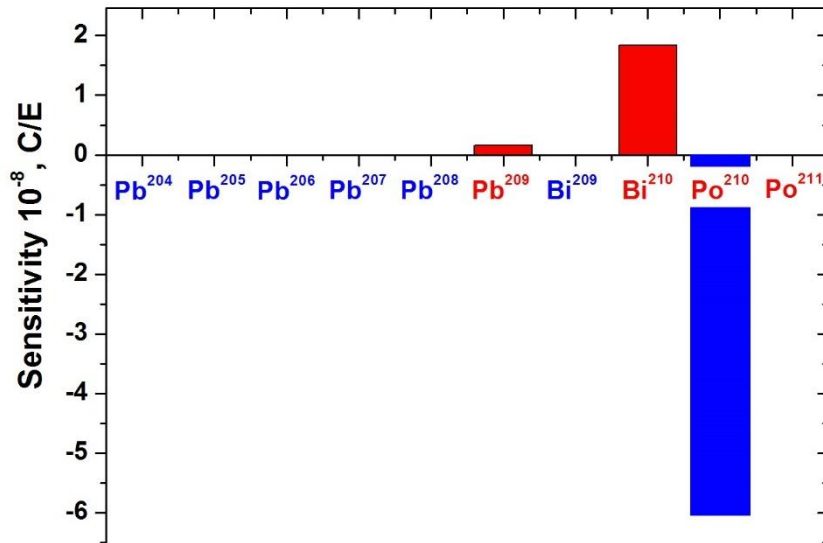


Sensitivities of the $Bi-209$ concentration to the (n,g) cross sections

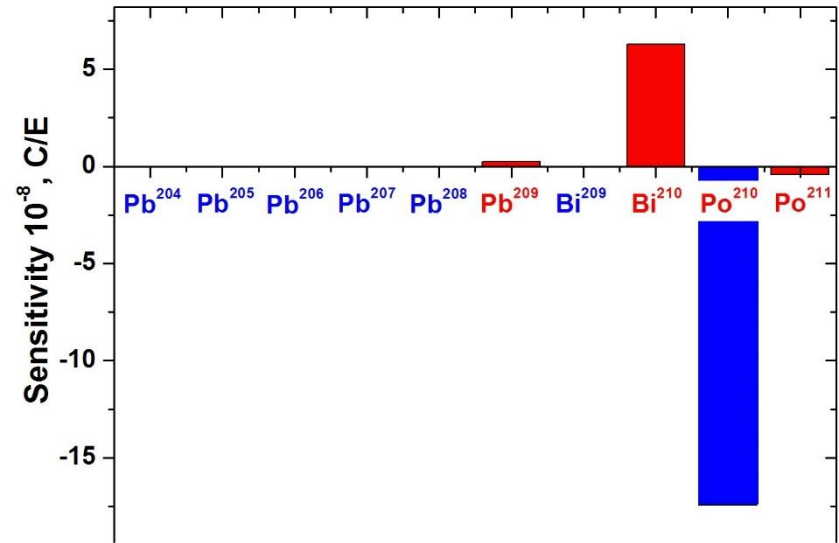
Sensitivity analysis of nuclear data

Also we can see which **decay data** is important to concentration changes of nuclides

$$S^i = \frac{dN_i(T)}{N_i(T)} / \frac{d\lambda}{\lambda} = \frac{\lambda}{N_i(T)} \cdot \int_0^T \left(N^* \frac{dM}{d\lambda} N \right) dt$$



Sensitivities of the **$Pb-208$** concentration to decay data



Sensitivities of the **$Bi-209$** concentration to decay data

Conclusion

Following results were obtained:

- The nuclide density of catalytic material in equilibrium state
- Heat density of catalytic material at several neutron flux level
- Sensitivity analysis of nuclides concentrations to CS and decay data

In the future decay heat from FPC of Molten Salt Reactor will be calculated using BurnUP equation.



THANK YOU

