

UNCLASSIFIED

Approved for Publication

This document is intended for publication in a journal, and is made available on the understanding that extracts or references will not be published prior to publication of the original, without the consent of the author.



United Kingdom Atomic Energy Authority

REACTOR GROUP

CHAPTER 6

FISSION PRODUCT ABSORPTION CROSS-SECTION

AT 2200 M/sec.

Thermal neutron absorption and resonance integrals

J. S. STORY

Industrial Power Reactors Division,

Atomic Energy Establishment,

Winfrith, Dorchester,

Dorset.



IAEA
NUCLEAR DATA UNIT
MASTER COPY

© UNITED KINGDOM ATOMIC ENERGY AUTHORITY, 1961

Enquiries about copyright and reproduction should be addressed
to the Winfrith Secretariat, Atomic Energy Establishment, Winfrith,
DORCHESTER, Dorset, England.

U.D.C.
539.17

UNCLASSIFIED

AEEW - M 122

CHAPTER 6

FISSION PRODUCT ABSORPTION CROSS-SECTIONS

AT 2200 M/sec.

Thermal neutron absorption and resonance integrals

by

J. S. STORY

ABSTRACT

This memorandum is to be published as one chapter (6) of a book entitled "Nuclear Data for Reactor Design, Vol I" by N. G. Sjostrand and J. S. Story. (publish Pergamon Press) and is a critical assessment of experimental data relating to the 2200 m/s. absorption cross-section of fission product isotopes.

Edited by A. L. POPE

A.E.E.,
Winfrith.

April, 1961.

W.3636

S.C. C.17

CONTENTS

	<u>Page</u>
6.1 Introduction	1
6.2 Experimental methods	2
6.3 Cross-section tables	3
Description and notation	
Corrections to experimental data	
Resonance integrals	
6.4 Short-lived fission products	5
6.5 The absorption cross-section of Xe^{135}	5
6.6 Recommendations for further work	12
References	15 to 17

TABLES

Table

6.1 Fission-product absorption of thermal and resonance neutrons	6
6.2 $^{54}Xe^{135}$; absorption cross-section at 2200 m/sec. from transmission measurements	9
6.3 $^{54}Xe^{135}$; absorption cross-section in various thermal reactor spectra	10

6.1 Introduction

Knowledge of the absorption cross-sections of fission-products is needed for calculating the long-term reactivity changes in nuclear reactors. Because they absorb neutrons, fission-products act as "poisons" by reducing the reactivity. The best known fission-product poisons are Xe^{135} and Sm^{149} , for these have very large absorption cross-sections and their effects are observable quite soon after start-up. However with continued irradiation of the fuel many other fission-products accumulate sufficiently to cause appreciable "poisoning".

Generally speaking only the stable or relatively long-lived fission-products accumulate sufficiently. To exemplify this statement consider the effects of a fission-product formed directly in fission during operation of a reactor at a steady flux ϕ . Admittedly the primary fission fragments are not usually significant poisons; however precursor nuclei are often sufficiently short-lived that a poison can be treated as if formed directly in fission. The saturation concentration of the fission product would be

$$N = \frac{y \cdot \Sigma_F \phi}{\lambda + \sigma_y \phi}, \quad (6.1)$$

where

y is the yield per fission of the fission product

λ is its decay constant

σ_y is its capture cross-section

Σ_F is the macroscopic fission cross-section of the fuel.

The poisoning P by a given nuclide may be defined as the ratio of the neutron absorption rate in the nuclide and the fission rate in the fuel, per unit volume. At saturation

$$P = \frac{y \sigma_y \phi}{\lambda + \sigma_y \phi} \quad (6.2)$$

P attains its maximum value at saturation concentration,

$$P = y,$$

if the nuclide is stable or long-lived ($\lambda \ll \sigma_y \phi$). Short-lived nuclei decay before they have time to absorb a neutron, and so have a smaller poisoning effect. For a numerical example suppose $\sigma_y = 100$ barns, and $\phi = 10^{14}$ neutrons/cm²sec. Then the product $\sigma_y \phi$ is 10^8 sec⁻¹ which corresponds to a half-life of about 3 years. Thus if the nuclide has a half-life of 1 month or less it has only a small chance of neutron capture before it decays.

Detailed assessments of the relative importance in thermal reactors of the various fission-product poisons have been given by PATTENDEN & ROSE (1958), PATTENDEN (1959), SHAFTMAN (1959), NEPHEW (1960), and WALKER (1960).

6.2 Experimental Methods

Several of the methods for cross-section measurement have been discussed in detail in chapter 3 (A.E.E. W-M57), and for the present chapter a brief outline will suffice. The following methods have been used to determine the capture cross-sections of fission products:

- A. Transmission measurements with neutron choppers and crystal spectrometers. The total cross-section is determined absolutely as a function of neutron energy in the low energy region. The absorption cross-section is obtained by subtracting the scattering cross-section. Since the latter is usually not known very accurately the method gives its best results when σ_A is much larger than σ_S . The method has been used by PATTENDEN (1958), and by SMITH et al. (1959).
- B. Pile oscillators The measurements are made by comparison with a standard absorber, and are most easily interpreted if the neutron spectrum has only a very small epithermal component. POMERANCE (1951, 1952) used a local oscillator in the reflector of a neutron reactor, and ROSE et al. (1958) used a reactivity oscillator in an internal thermal column or "thermal well". Gold and boron respectively were used as standard absorbers.
- It should be noted that in the local oscillator used by POMERANCE, the neutron counter has approximately a $1/v$ response to the absorbed neutrons. There are two consequences:
- (i) The oscillator operates in a region where the epithermal neutron flux is very small, and the counter response practically annuls any residual epithermal effect.
 - (ii) None-the-less the average cross-section measured is not strictly $g \sigma_0$ unless $g = 1$. It is more nearly σ_0 , unless there is a resonance in the thermal region.
- C. Activation. This method is of great value when the samples available are very small. The samples are usually irradiated in a thermal column or inside a reactor, and cadmium-difference techniques may be employed to simplify the interpretation. The measurements are made in comparison with an activation standard used to monitor the neutron flux. For reliable results it is necessary to have a fairly detailed understanding of the decay characteristics of the various activities observed, and particularly for isomeric activities. Many of the older measurements are unreliable in these respects. A classical survey by this method was made by SEREN et al. (1947) who give an admirable description of the difficulties encountered. An example of more recent applications is the work of EASTWOOD et al. (1958).
- D. Mass spectrometry. If the cross-sections are fairly large, prolonged irradiation may cause observable changes in the mass spectrum. Usually the sample is irradiated in a reactor. Since a large capture cross-section almost always implies that there is a resonance at low neutron energies the measurements cannot always be interpreted in a very satisfactory way. However the method does permit the identification of strong "poisons", and has the added virtue that only small samples are needed. Recent work has been reported by ANIKINA et al. (1958) and by BIDINOSTI et al. (1958).

Broadly speaking one expects the most reliable data from the pile oscillator method, or from the transmission method if σ_A is large; however errors may ensue if the samples contain impurities. For small cross-sections (<0.5 barn) activation methods may be more reliable, and pure samples are usually not necessary. The mass spectrometer method has permitted the identification of many strong absorbers, particularly among those leading to stable product nuclides; but it seems that very careful work is needed for reliable results.

6.3 Cross-section tables

Description and notation

In Tables 6.1, 6.2 and 6.3 following, are given the cross-sections for all those fission-products listed by KATCOFF (1958) for which the data are available, and for a few additional nuclei also, which may be formed through neutron capture in fission-product nuclie.

The elements are listed in table 6.1 in order of increasing atomic number, except for Xe¹³⁵. This important nuclide is treated separately in section 6.5 and in tables 6.2 and 6.3, which follow table 6.1. The following notation is used:

σ_0 denotes an absorption or activation cross-section for 2200m/sec neutrons.

$\hat{\sigma}$ denotes the effective cross-section in the experimental spectrum,

$$\hat{\sigma} = (g + rs)\sigma_0$$

in the terminology of WESTCOTT (1958), see chapter 1.

Σ'_r denotes the resonance absorption integral for dilute material, excluding the $1/v$ contribution.

$$\Sigma'_r = \int_{\mu kT}^{\infty} \{\sigma(E) - \sigma_0/(E_0/E)\} dE/E$$

$$= \sigma_0 \left\{ \frac{\sqrt{\pi}}{2} s + 2(g-1) \sqrt{\mu}/(T_0/T) \right\},$$

the notations of WESTCOTT (1958) being used on the right.

* denotes that probably the cross-section obeys the $1/v$ law approximately in the thermal region, so that $g \approx 1.00$

X denotes that g and s coefficients have been tabulated, the cross-section having been measured as a function of neutron energy.

The column headed "Reaction" shows whether the data are for total neutron absorption by the nuclide, or it gives the half-life of the product activity if an activation measurement was made.

Recommended values are underlined or are mentioned explicitly.

r_o denotes $\nu(T_o/T)$, in the terminology of WESTCOTT (1958).

Most of the data listed in table 6.1 were measured in broad neutron spectra, by methods B, C, or D, of section 6.2. Values for σ_o , the 2200m/sec cross-section, cannot be derived unless the neutron spectrum and the energy variation of the cross-section are known. However for many nuclides it is reasonable to presume that the capture cross-sections obey the $1/v$ law in the thermal region at least approximately; then

$$g \approx 1.00$$

and for measurements in thermal spectra with $r \approx 0$, or in sub-cadmium spectra,

$$\sigma_o = \delta$$

Corrections to experimental data

It may be noticed that many of the data differ slightly from the values given in the references cited. It has been possible to revise a good many of the comparative measurements, using the values recommended in chapter 3 for the standard cross-sections at 2200m/sec. For example, the data of SMITH & REEDER (1955, 1956), ROY and co-workers (1956-1958), BAERG et al. (1958), BIDINOSTI et al. (1958), KENNETT & THODE (1958), FICKEL & TOMLINSON (1959), have been revised using a half-life of 5.23 years for Co^{60} and $\sigma_{\text{act}}[\text{Co}] = 37.7$ barns at 2200m/sec. The data of BOTHE (1946) have been amended using $\delta_{\text{act}}[\text{Ho}] = 59.6$ barns.

It is not generally feasible to amend activation data for improved half-life values, unless it can be assumed that the irradiation and cooling periods were much shorter than the half-life. However some of the activation cross-sections reported by SEREN et al. (1947) have been revised by using more up-to-date values for the relative isotopic abundances in natural samples

Resonance integrals

Resonance integrals have been included for convenience. We have not attempted to present all resonance integral data for each nuclide, but have aimed to give representative values. The data of HARRIS et al. (1950a) appear to be too low in general, presumably because the samples were not sufficiently dilute. TATTERSALL et al. (1960) on the other hand may have over-compensated for resonance self-shielding. We tend to prefer the values calculated from resonance parameters. WALKER (1960) seems usually to have used the most recent data.

⁷ For example, transmission measurements with a sample of natural isotopic composition may show $1/v$ behaviour in the thermal region (see BNL 325). Then it is reasonable to guess that the individual isotopes would give similar results. This inference is weakened when applied to an isotope of low relative abundance.

The resonance integral data also have been revised as described in the preceding paragraph. For example the data of KLIMENTOV & GRYAZEV (1957) have been renormalised so that they give approximately the expected results for both boron and uranium.

6.4 Short-lived fission-products

Besides the information given in tables 6.1, 6.2 and 6.3, qualitative information has been obtained by mass-spectrometry for several fission-product chains. BIDINOSTI et al. (1958) pointed out that the effect of neutron capture by the short-lived fission-products will be most apparent if the fissile sample is irradiated in a very high neutron flux, and would result in unexpected changes in the yields of the stable and long-lived end products. Using this principle they were able to show that there is no significant neutron absorption amongst the short-lived precursors of the 23 chains with the following mass-numbers

88	-	90
92	-	96
133	-	137
	142	
144	-	152

6.5 The absorption cross-section of Xe¹³⁵

In thermal reactors Xe¹³⁵ is the most important of the fission-product poisons, because of both the extremely large thermal neutron cross-section $\sim 3 \times 10^6$ barns and the short half-life of 9.18 hours. The thermal cross-section has nearly the maximum possible value $\pi g^* \lambda^2$, in consequence of a low energy resonance with an unusually large neutron reduced-width; $E_{res} = 0.084$ eV and $g^* T_n^0 \approx 0.013$ eV. The existence of this resonance is the more noteworthy in that the compound nucleus Xe¹³⁶ is a "closed shell" nucleus with 82 neutrons. Consequently the resonance spacing is particularly wide, ~ 600 eV in the low energy region.

With a radioactive nuclide of such a short half-life it is a formidable task to accrete, extract, and handle a sample large enough for measurements with a neutron spectrometer. In spite of these difficulties the total cross-section has been measured as a function of neutron energy, first by BERNSTEIN et al. (1956)* who used a bent-crystal focussing spectrometer, and more recently by SMITH et al. (1959)* with a fast neutron-chopper. A more copious supply of fission-product Xe was available to SMITH et al., so they had the benefits of a much larger sample.

The cross-section determined by BERNSTEIN et al. is systematically lower than the data of SMITH et al. At thermal energies the difference is of the order of 12%. This is significantly greater than would be expected from the uncertainties assigned to the data, but is not very surprising for so difficult a measurement.

* The work was carried out in 1948 and 1954, respectively.

TABLE 6.1

Fission-product absorption of thermal and resonance neutrons

Ge
32

Reference	Mass no.	Half-life	$\hat{\sigma}$ barns	σ_0 barns	Σ_r barns	Reaction	Method and comments
POLTERANCE (1952)	72	Stable		0.97 ± 0.10 *	Small	Absorption	Local oscillator, $r \approx 0$ No resonance known below 4 keV
POLTERANCE (1952)	73	Stable		14 ± 1 *	Small	Absorption	Local oscillator, $r \approx 0$ No resonance known below 100 eV
SERENI et al (1947) REYNOLDS (1950a)	74	Stable	0.38 ± 0.08 0.53 ± 0.10	*		\rightarrow 32 min \rightarrow 32 min	Activation in reactor Activation with sub-Cd neutrons
POLTERANCE (1952) FLAMMERSFELD (1952)			$0.43 \times \hat{\sigma} [\rightarrow 82 \text{ m}]$ $\left\{ \begin{array}{l} 0.042 \pm 0.008 \\ 0.22 \pm 0.04 \end{array} \right.$	0.62 ± 0.06 $\left\{ \begin{array}{l} * \\ * \end{array} \right.$		Absorption \rightarrow 48 sec \rightarrow 48 sec \rightarrow 82 min ?	Activation Activation with sub-Cd neutrons } Probably from activation cadmium-ratio
der MATEOSIAN & GOLDHABER (1957) SUNYAR & GOLDHABER (1949)				≤ 10	Small	Absorption	No resonance known below 370 ev

The 48 sec isomeric state of Ge 75 decays by isomeric transition to the ground state. The values given for $\hat{\sigma}$ act [$\rightarrow 82$ min] are for production of the ground state both directly, and by complete decay of the short-lived isomer.

Table 6.132^{Ge}continued

References	Mass no.	Half-life	$\hat{\sigma}$ barns	σ_0 barns	Σ'_{r} barns	Reaction	Method and Comments
SEREN et al. (1947)	76	Stable	0.071 ± 0.014 $1.1 \times \hat{\sigma} [\rightarrow 11 \text{ hr}]$	*	*	$\rightarrow 11 \text{ hr}$ $\rightarrow 54 \text{ sec}$	Activation in reactor Activation in reactor
ARNOLD & SUGARMAN (1947)							
REYNOLDS (1950a)			$\left\{ \begin{array}{l} 0.015 \pm 0.006 \\ 0.30 \pm 0.06 \end{array} \right.$	*		$\rightarrow 54 \text{ sec } \beta^-$ $\rightarrow 11 \text{ hr}$	Activation with sub-Cd neutrons
POLLERANCE (1952)				0.36 ± 0.07			Local oscillator; $r \approx 0$
LYON & ELDRIDGE (1957)			$\left\{ \begin{array}{l} 0.10 \pm 0.01 \\ 0.043 \pm 0.002 \end{array} \right.$	*		$\rightarrow 54 \text{ sec } \beta^-$ $\rightarrow 11 \text{ hr}$	Absorption
der MATEOSIAN & GOLDHABER (1957)			$\left\{ \begin{array}{l} 0.076 \pm 0.015 \\ 0.089 \pm 0.015 \end{array} \right.$	*		$\rightarrow 54 \text{ sec } \beta^-$ $\rightarrow 11 \text{ hr}$	Absorption
					Small		No resonance known below 550 eV

The 54 sec isomeric of Ge⁷⁷ decays ~ 30% by isomeric transitions to the ground state, and the rest by beta emission to As⁷⁷. The values given here for $\hat{\sigma}$ act [$\rightarrow 54 \text{ sec } \beta^-$] are for production of the ~ 70% beta activity only. The values for $\hat{\sigma}$ act [$\rightarrow 11 \text{ hr}$] are for production of the ground state both directly and by decay of the short-lived isomer.

Table 6.1

33^{A8}

References	Mass no.	Half-life	$\hat{\sigma}$ barns	σ_0 barns	$Z' r$ barns	Reaction	Method and comments
SEREN et al. (1947)	75	Stable	4.2 ± 0.8	*		$\rightarrow 26.4$ hr	Activation in thermal column
COLLER & LITTLER (1950)			5.2 ± 0.3	*		Absorption	Reactivity oscillator in reactor spectrum
HARRIS et al. (1950b)			5.8	*		Absorption	Reactivity oscillator; $r_0 = 0.046$
POMERANCE (1951)				<u>4.29 ± 0.22</u>		Absorption	Local oscillator; $r \approx 0$
HARRIS et al. (1950a)					36	$\rightarrow 26.4$ hr	Activation Cd ratio. Revised

Table 6.1

34.30

Reference	Mass no.	Half-life	$\hat{\sigma}$ barns	σ_0 barns	Σ'_{r} barns	Reaction	Method and comments
ARNOLD & SUGARMAN (1947)	76	Stable	≈ 6	*	85 ± 7	$\rightarrow 17$ sec Absorption	Activation of isomeric state, in reactor Local oscillator; $r \approx 0$ No resonance known below 860 eV
POMERANCE (1952)					Small		
POMERANCE (1952) WALKER (1960)	77	Stable		41 ± 4 *	≈ 12	Absorption Absorption	Local oscillator; $r \approx 0$ Calculated from resonance parameters
POMERANCE (1952)	78	Stable		0.4 ± 0.4 *	Small	Absorption	Local oscillator; $r \approx 0$ No resonance known below 380 eV
SEREN et al. (1947) POMERANCE (1952) WALKER (1960)	80	Stable	0.034 ± 0.007 0.46 ± 0.09	*	0.61 ± 0.06	$\rightarrow 57$ min $\rightarrow 18$ min Absorption	Activation in reactor Activation in thermal column Local oscillator; $r \approx 0$ Calculated from resonance parameters
SEREN et al. (1947) ARNOLD & SUGARMAN (1947) POMERANCE (1952)	82	Stable	0.063 ± 0.013 $\frac{0.044}{0.0041}$	*	≈ 0.6	$\rightarrow 70$ sec $\rightarrow 26$ min $\rightarrow 70$ sec $\rightarrow 26$ min Absorption	Activation in reactor. Activity of Br^{83} daughter observed) Activation in reactor. Based on σ [nat. Se \rightarrow 18 min Se^{81}] = 0.23 barns Local oscillator; $r \approx 0$ No resonance known below 3 keV

Table 6.1

 ^{75}Br

Reference	Mass no.	Half-life	$\hat{\sigma}$ barns	σ_0 barns	$\Sigma' r$ barns	Reaction	Method and comments
SEREN et al. (1947)	79	Stable	2.9 ± 0.6 8.5 ± 1.7	*	*	$\rightarrow 4.4$ hr $\rightarrow 18$ min	{ Activation in thermal column. Corrected for 5% decay for Br-80 ground state by orbital electron capture
REYNOLDS (1950b)			10.3 ± 0.8	*		Absorption	Mass spectrometry with isotope dilution. Result is for thermal neutrons see footnote
HARRIS et al. (1950a)						$\rightarrow ?$	Activation Cd-ratios. Recalculated
KLIMENTOV & GRYAZEV (1957)						Absorption	Reactivity measurements with natural Br under Cd. Recalculated, and Br-81 contribution subtracted
Authors (1960)						Absorption	Calculated from resonance parameters
SEREN et al. (1947)	81	Stable	2.3 ± 0.5	*		$\rightarrow 36$ hr	Activation in reactor
REYNOLDS (1950b)			2.8 ± 0.2	*		Absorption	As for Br-79, above
ORNL unpublished work quoted in BNL 325 (1960)			3.3 ± 0.4	*		$\rightarrow 36$ hr	Activation
WALKER (1960)					≈ 60	Absorption	Calculated from resonance parameters

Note: REYNOLDS (1950b) measured the ratio $\sigma_A [Br^{79}] / \sigma_A [Br^{81}] = 3.86 \pm 0.05$ in a reactor spectrum. We have combined this with σ_A [natural Br] = 6.6 ± 0.1 barns at 2200 m/sec from the work of POMERANCE (1951) and EGELSTAFF (1953), and with the resonance integrals. Both isotopes have nearly the same value for $\Sigma' r / \sigma_0$ so that $\hat{\sigma}$ for thermal neutrons can be calculated with little uncertainty.

Table 6.1

 ^{36}Kr

Reference	Mass no.	Half-life	$\frac{\sigma}{\sigma_0}$ barns	Σ_r' barns	Reaction	Method and comments
MAGNAMARA & THODE (1950)	82	Stable	45 ± 15	*	(n, γ)	Mass spectrometry. Reactor spectrum. Revised uncertainty from BNL 325 (1960)
WALKER (1960)				~180	Absorption	Calculated from resonance parameters
MAGNAMARA & THODE (1950)	83	Stable	205 ± 30	*	(n, γ)	Mass spectrometry. Reactor spectrum. Revised uncertainty from BNL 325 (1960)
KENNEDY & THODE (1958)			227 ± 45	*	(n, γ)	Mass spectrometry. Reactor spectrum. Recalculated
WALKER (1960)				~140	Absorption	Calculated from resonance parameters
HOAGLAND & SUGARMAN (1945)	84	Stable	0.10 ± 0.03 0.06 ± 0.02	**	$\rightarrow 4.4 \text{ hr } \beta^-$ - 10 yr	Activation in reactor
MAGNAMARA & THODE (1950)			< 2		(n, γ)	Mass spectrometry. Reactor spectrum
				~3.3		Calculated from resonance parameters
MAGNAMARA & THODE (1950)	85	10.5 yr	<15		(n, γ)	Mass spectrometry. Reactor spectrum
SUGARMAN & TURKEVICH (1944)	86	Stable	0.06	*	→76 min	Activation in reactor
HOAGLAND & SUGARMAN (1945)			0.06 ± 0.02	*	→ 76 min	Activation in reactor
MAGNAMARA & THODE (1950)			< 2	*	(n, γ)	Mass spectrometry. Reactor spectrum
SUGARMAN & TURKEVICH (1944)	87	76 min	<600		→ 2.3 hr	Activation in reactor. Original value ≤ 470 barns, revised in BNL 325 (1958)

The 4.4 hr isomeric state of Kr^{85} decays ~2% by isomeric transitions to the 10 yr ground state, and the remainder transforms to Rb^{87} by beta emission. The value given for $\sigma_{act} [\rightarrow 4.4 \text{ hr } \beta^-]$ is for production of the ~7% beta activity alone. The value for $\sigma_{act} [\rightarrow 10 \text{ yr}]$ is for production of the 10 yr ground state, both directly and by decay of the short-lived isomer

Table 6.1

 ^{37}Rb

Reference	Mass no.	Half-life	$\hat{\sigma}$ barns	σ_0 barns	Σ'_{r} barns	Reaction	Method and comments
SEREN et al. (1947)	85	Stable	0.72 ± 0.15	*		$\rightarrow 18.7$ day	Activation in reactor
POMERANCE (1951)				0.96 ± 0.10		Absorption	Local oscillator; $r \approx 0.87$
ORNL unpublished work quoted in BNL 325 (1960)			0.91 ± 0.08	*		$\rightarrow 18.7$ day	Subtraction of $\sigma_{\text{act}}[\text{Rb}^{87}]$ from σ_A [natural Rb]
WALKER (1960)							Activation
		Recommended value		<u>0.93 ± 0.08</u>			Calculated from resonance parameters
SEREN et al. (1947)	87	5×10^{-10} yr	0.118 ± 0.024	*		$\rightarrow 18$ min	Activation in thermal column
WALKER (1960)					≈ 0.11		Calculated from resonance parameters
ROY et al. (1958)	88	18 min	1.08 ± 0.32			$\rightarrow 15$ min	Activation in reactor; $r_0 \approx 0.02$. Revised

Table 6.1

 ^{38}Sr

Reference	Mass no.	Half-life	$\frac{\sigma}{\text{barns}}$	σ_0 barns	Σ^r barns	Reaction	Method and comments
SEREN et al. (1947) ORNL unpublished work quoted in BNL 325 (1960)	86	Stable	1.31 ± 0.26	*		$\rightarrow 2.8$ hr	Activation of isomer in thermal column
POWERANCE (1951) KLEMENTOV & GRYAZEV (1957)			1.65 ± 0.16	*		$\rightarrow 2.8$ hr	Activation of isomer
POWERANCE (1951) WALKER (1960)	87	Stable		≤ 10		Absorption	Local oscillator; $r \approx 0$ Subtraction of known activation cross-sections from σ_A [natural Sr]
SEREN et al. (1947) ROY et al. (1958)	88	Stable			≤ 50	Absorption	Reactivity measurements with natural Sr under Cd. Recalculated and contributions of Sr^{87} and Sr^{88} subtracted
POWERANCE (1951) WALKER (1960)						Absorption	As above
ROY & ROY (1957) ANL unpublished work quoted in BNL 325 (1955)	89	50 day	0.43 ± 0.04			$\rightarrow 28$ day	Activation in reactor with $r_0 \approx 0.02$. Recalculated, using revised input data for Co and Sr 88
	90	28 yr	1.0 ± 0.6			$\rightarrow 9.7$ hr	Activation

There is evidence for an isomeric state of Sr^{89} of 10 days half-life decaying by isomeric transition to the 50 day ground state. The values given for $\sigma_{\text{act}} [\rightarrow 50 \text{ day}]$ are for production of the 50 day activity and probably include the production via the isomeric state.

For Sr^{86} and Sr^{87} only upper limits have been given for σ_A , but these values are correlated. If the maximum value holds for one of the two isotopes, then the other can have only a small cross-section.

Table 6.1

397

Reference	Mass no.	Half-life	$\hat{\sigma}$ barns	σ_0 barns	$\Sigma' r$ barns	Reaction	Method and comment
BOTHE (1946)	89	Stable	1.2 ± 0.3	*		$\rightarrow 64$ hr	Activation
SERENI et al. (1947)			1.24 ± 0.25	*		$\rightarrow 64$ hr	Activation in reactor
PICERANCE (1951)				1.43 ± 0.14		Absorption	Local oscillator; $r \approx 0$
BERNOIS et al. (1951)			1.33 ± 0.19	*		Absorption	Reactivity oscillator in reactor spectrum. $r_0 = 0.03$
ORNL unpublished work quoted in BNL 325 (1957, 1958)			1.31 ± 0.08	*		Absorption	
- ditto -			1.26 ± 0.08	*		$\rightarrow 64$ hr	Activation
HARRIS et al. (1950a)					0.27	$\rightarrow 64$ hr	Activation Cd ratio
WALKER (1960)					≈ 0.1	Absorption	Calculated from resonance parameters
				1.3 ± 0.1			
SMITH & REEDER (1955)	90	64 hr	<7.3			$\rightarrow 58$ day	Activation; $r = 0$ assumed. Includes the production of 58 day Y_{91} by decay of the 50 min. isomer. Revised
EASTWOOD et al. (1958)	91	58 day	<10			$\rightarrow 3.6$ hr	Activation in reactor; $r_0 \approx 0.02$
Chalk River unpublished work quoted in BNL 325 (1960)			1.07 ± 0.09			$\rightarrow 3.6$ hr	Activation

Table 6.1

^{40}Zr

References	Mass no.	Half -life	$\hat{\sigma}$ barns	σ_0 barns	Σ_r barns	Reaction	Method and comments
POLERANCE (1952)	90	Stable		0.11 ± 0.02 *	Small	Absorption	Local oscillator; $r \approx 0$ No resonance known below 300 eV
POLERANCE (1952) WALKER (1960)	91	Stable		1.57 ± 0.13 *	~ 4.8	Absorption	Local oscillator; $r \approx 0$ Calculated from resonance parameters
POLERANCE (1952)	92	Stable		0.26 ± 0.08 *	Small	Absorption	Local oscillator; $r \approx 0$ No resonance known below 300 eV
POLERANCE (1955)	93	1.1×10^6 y		< 4 *		Absorption	Local oscillator; $r \approx 0$. No thermal resonance assumed
SEREN et al. (1947) LEDDICOTTE & REYNOLDS (1949)	94	Stable	0.42 \pm 0.18	*		$\rightarrow 65$ day	Activation in reactor
POLERANCE (1952) ORNL unpublished work quoted in BNL 225 (1960)			0.060 ± 0.01 0.076 ± 0.008	0.080 ± 0.040 *		$\rightarrow 65$ day Absorption $\rightarrow 65$ day	Activation with sub-Cd neutrons Local oscillator; $r \approx 0$ Activation No resonance known below 300

Table 6.140^{Zr}continued

Reference	Mass no.	Half-life	σ barns	σ_0 barns	Σ'_{R} barns	Reaction	Method and comments
SEREN et al. (1947)	96	Stable	0.32 ± 0.06			$\rightarrow 17$ hr	Activation in reactor
LEDDICOTTE (1949)			0.045			$\rightarrow 17$ hr	Activation
PONERANCE (1952)				0.10 ± 0.10		Absorption	Local oscillator; $r \approx 0$
CERN unpublished work quoted in BNL 325 (1960)			0.053 ± 0.005			$\rightarrow 17$ hr	Activation
SUKYAR & GOLDHABER (1949)					$\lesssim 10$?	Probably from activation cadmium-ratio

Revised values of $\sigma_{\text{act}}[\text{Zr}^{94}]$ and $\sigma_{\text{act}}[\text{Zr}^{96}]$, obtained by re-evaluation of the early data, have been listed without explanation in BNL 325 (1955, 1957 and 1958). Presumably these revisions were necessitated by discovery of the complex decay schemes of Zr⁹⁵ and Zr⁹⁷ with their Nb daughters in equilibrium. This serves but to emphasise the unreliability of the early activation data, and the need for new measurements.

Table 6.1

Nb_{41}

Reference	Mass no.	Half-life	Δ barns	σ_0 barns	Σ_F' barns	Reaction	Method and comments
SZEN et al (1947)	93	Stable	1.0 ± 0.4	*		\rightarrow 6.6 min	Activation in thermal column of short lived isomer of Nb_{94}
ANDERSON et al (1944)			1.55			Absorption	Reactivity; reactor spectrum
COLLER & LITTLE (1950)			1.35 ± 0.15	*		Absorption	Reactivity oscillator; in reactor spectrum
HARRIS et al (1950b)			1.61	*		Absorption	Ditto, with $r_c = 0.046$
POLTRANCE (1951)			1.17 ± 0.02	1.1 ± 0.1		Absorption	Local oscillator; $r \approx 0$, ($g \approx 1.0$ assumed)
ROSE et al (1959)				1.17 ± 0.02		Absorption	Reactivity oscillator in thermal well; $r \approx 0$, ($g = 1.000$ assumed)
TATTERSAL et al (1960)					13 ± 5	Absorption	Reactivity oscillator in thermal well and lattice spectra
- ditto -					15 ± 5	Absorption	Calculated from resonance parameters

Table 3.1

 ^{42}Mo

Reference	Mass no.	Half-life	$\frac{\Delta}{\text{barns}}$	σ_0 barns	Σ_r^1 barns	Reaction	Method and comments
POLERACE (1952)	95	Stable		13.9 ± 1.1		Absorption	Local oscillator; $r \approx 0$
TATTERSALL et al. (1960)			14.7 ± 0.5	*		Absorption	Reactivity oscillator in thermal well; $r \approx 0$
- ditto -					100 ± 20	Absorption	Ditto, with lattice spectrum also
- ditto -					100 ± 15	Absorption	Calculated from resonance parameters
				Recommended value 14.5 ± 0.5			
POLERACE (1952)	96	Stable		1.2 ± 0.6		Absorption	Local oscillator; $r \approx 0$
WALKER (1960)					~ 26	Absorption	Calculated from resonance parameters
POLERACE (1952)	97	Stable		2.2 ± 0.7		Absorption	Local oscillator; $r \approx 0$
WALKER (1960)					14 ± 3	Absorption	Calculated from resonance parameters
SEZEN et al. (1947)	98	Stable	0.42 ± 0.08	*		\rightarrow 67 hr	Activation in reactor
POLERACE (1952)			0.21 ± 0.03	*		Absorption	Local oscillator; $r \approx 0$
ANDERS (1958)			0.51 ± 0.06	*		\rightarrow 67 hr	Activation; $r \approx 0$
ORNL unpublished work quoted in EIL 325 (1960)						\rightarrow 67 hr	Activation
ANDERS (1958)					13 ± 5	\rightarrow 67 hr	Activation in different spectra with Cd-ratios for Au
WALKER (1960)					~ 3.4	Absorption	Calculated from resonance parameters
SEZEN et al. (1947)	100	Stable	0.46 ± 0.09			\rightarrow 14.6 min	Activation in reactor
POLERACE (1952)				0.5 ± 0.5		Absorption	Local oscillator; $r \approx 0$
Unpublished work quoted in EIL 325 (1955)			0.22 ± 0.05			\rightarrow 14.6 min	Activation
CARELL (1960)			0.197 ± 0.008			\rightarrow 14.6 min	Activation with sub-Cd neutrons
- ditto -					3.1 ± 0.2	\rightarrow 14.6 min	Activation Cd-ratios
WALKER (1960)				$\gtrsim 0.4$	~ 6.2		Calculated from resonance parameters

Table 6.1

^{43}Tr

Reference	Mass no.	Half-life	$\frac{\sigma}{\sigma_0}$ barns	Σ_r^1 barns	Reaction	Method and comments
POLARANCE (1955b)	99	2.1×10^5 y			Absorption	Local oscillator; $r \approx 0$
PATTERDEN (1958)					Absorption	Transmission, using a crystal spectrometer
- ditto -					Absorption	Tables of σ and s coefficients are given
TATTERSALL et al. (1960)					Absorption	Reactivity oscillators in reactor spectra
WALKER (1960)					Absorption	Results given here are for σ (thermal), σ_0 , and Σ_r . $\sigma = 1.009$ is assumed.
					Absorption	Calculated from resonance parameters
					Recommended value	22 ± 3

Table 6.1

Ru

Reference	Mass no.	Half-life	$\bar{\sigma}$ barns	Σ^*_r barns	Reaction	Method and comments
POLERANCE (1951)	99	Stable	* ≤ 15		Absorption	Local oscillator, with natural Ru, and subtraction of known activation cross-sections. $r \approx 0$
NEPHEW (1960)				≤ 350	Absorption	Calculated from resonance parameters of natural Ru
POLERANCE (1951)	100	Stable	* ≤ 15		Absorption	As for Ru ⁹⁹
NEPHEW (1960)			*	≤ 350	Absorption	As for Ru ⁹⁹
POLERANCE (1951)	101	Stable		≤ 12	Absorption	As for Ru ⁹⁹
NEPHEW (1960)				≤ 260	Absorption	As for Ru ⁹⁹
SERENI et al. (1947)	102	Stable	1.17 ± 0.25	*	→	39.8 day Activation in reactor
KATCOFF & WILLIAMS (1951) - ditto -			1.44 ± 0.16	*	→	39.8 day Activation with sub-Cd neutrons
SERENI et al (1947)	104	Stable	0.74 ± 0.15	*	4.0 → 39.8 day	Calculated from Cd-ratios
SUNYAR & GOLDHABER (1949)				≤ 10	?	Activation in reactor. Includes data from activation of daughter, 37 hr Rh ¹⁰⁵
SHARMA & POOL (1959)	105	4.5 hr	0.20 ± 0.06		→ 1.0 yr	Probably from activation Cd-ratio Activation, probably in reactor spectrum

The upper limits given for σ_0 of Ru⁹⁹ are correlated. If σ_0 has its maximum value for one of these isotopes then it must be very small for the other two. Similar remarks apply to the values for Σ^*_r for these isotopes, (but not for Ru¹⁰⁴)

Table 6.1 ^{45}Rh

Reference	Mass no.	Half-life	$\frac{\Delta}{\sigma}$ barns	σ_{abs} barns	Σ^r barns	Reaction	Method and comments
SEREN et al. (1947)	103	Stable	11.6 ± 1.2			\rightarrow 4.4 min \rightarrow 44 sec	{ Activation in thermal column
KELJER (1949)			137 ± 14				
HARRIS et al. (1950b)		185		140 ± 3			Transmission using slow chopper
POLERANCE (1951)							Reactivity oscillator in reactor's pectrum; $r_0 = 0.046$
SAILOR (1953)				156 ± 8			Local oscillator; $r \approx 0$
ROSE et al. (1958)				142 ± 10			Absorption
HARRIS et al. (1950a)		146 \pm 5					Absorption
WALKER (1960)							Absorption
WESTCOTT (1960)							Absorption
							Reactivity oscillator in thermal well; $r \approx 0$, $\epsilon = 1.023$ at 200°C
							Activation Cd-ratios. Revised Calculated from resonance parameters
							Tables of ϵ and s coefficient
							X Recommended value 143 ± 5
							$\frac{1020 \pm 100}{1020 \pm 20} \rightarrow ?$
							$\frac{1020 \pm 100}{1020 \pm 5} \rightarrow$
							The 4.4 min isomeric state of Rh^{104} decays almost entirely by isomeric transition to the 44 sec ground state. The activation cross-sections given are for the independent direct production of the two isomers through neutron capture in Rh^{103} .
Chalk River unpublished work quoted in BNL 225 (1960)	104	4.4 min 44 sec	300 \pm 100 40 \pm 30			\rightarrow 36 hr \rightarrow 36 hr	Activation in reactor - ditto -

Table 6.1

 ^{46}Pd

Reference	Mass no.	Half-life	σ barns	$\Sigma \sigma$ barns	Reaction	Method and comments
POLTRANCE (1951)	104	Stable		≤ 4.0	Absorption	Local oscillator; $r \approx 0$. Subtraction of known activation cross-sections from $\sigma_A / \text{natural Pd}$
WALKER (1960)				≥ 0.5	~ 1.4	Calculated from resonance parameters
POLTRANCE (1951)	105	Stable		≤ 17	Absorption	As for Pa^{104} above
WALKER (1960)				≥ 1.35	≈ 67	Calculated from resonance parameters
POLTRANCE (1951)	106	Stable		≤ 14	Absorption	As for Pa^{104} above
WALKER (1960)				≥ 0.2	~ 5.5	Calculated from resonance parameters
SEEBEL et al. (1947)	106	Stable	11.2 ± 2.2		$\rightarrow 13.5$ hr	Activation in thermal column
LEISTER (1958)			9.3 ± 0.7		$\rightarrow 13.5$ hr	Activation; $r \approx 0$ assumed
SEIGAL et al. (1959)			0.26 ± 0.04		$\rightarrow 4.8$ min	Activation in thermal column
WALKER (1960)			14.4 ± 2.1		$\rightarrow 13.5$ hr	Recalculated with $\sigma_{\text{Au}} = 99.0$ b
				≥ 4.7	≈ 150	Calculated from resonance parameters
The 4.8 min isomeric state of Pd^{109} decays by isomeric transition to the ground state. The values given here for σ_{act} ($\rightarrow 13.5$ hr) are for production of the 13.5 hr ground state both directly and by decay of the short-lived isomer.						
SEEBEL et al. (1947)	110	Stable	0.39 ± 0.08		$\rightarrow 23.6$ min	Activation in thermal column
SEIGAL et al. (1959)			< 0.05		$\rightarrow 5.5$ hr) Activation in thermal column
WALKER (1960)			0.19 ± 0.03		$\rightarrow 23.6$ min) Calculated from resonance parameters
				~ 6.0		

Upper limits are given for the thermal cross-sections of Pd^{104} , Pd^{105} and Pd^{106} . These values are correlated, so that if the maximum value holds for one isotope then the cross-sections of the other two must be close to zero.

Table 6.1

 ^{147}Ag

Reference	Mass no.	Half-life	σ barns	Σ'_{fr} barns	Reaction	Method and results
See chapter 3	107	Stable	34.9 ± 2.5	69	$\rightarrow 2.3$ min.	Activation Cd ratio. Recalculated
HARRIS et al. (1950a)				25	Absorption	Calculated from resonance parameters
MATTERSALL et al. (1960)						
See chapter 3	109	Stable	92.4 ± 2.7		Absorption	
- ditto -			0.9 ± 0.6		$\rightarrow 253$ day	Activation
- ditto -			91.5 ± 2.8		$\rightarrow 24$ sec.)
HARRIS et al. (1950a)				1240	$\rightarrow ?$	Activation Cd ratios. Recalculated
MATTERSALL et al. (1960)						Reactivity oscillator in thermal well and lattice spectra
- ditto -				1870 \pm 200	Absorption	
WESTCOTT (1960)				1400 ± 65	Absorption	Calculated from resonance parameters. Mainly due to resonance at 5.2 eV
					X	Table of g-coefficients for natural Ag^{107} . The deviation from unity is due to Ag^{109}

Reference	Mass no.	Half-life	σ barns	Σ_F barns	Reaction	Method and comments
Unpublished work (1948) quoted in BNL 325	110	Stable	0.2 ± 0.1	*	$\rightarrow 49$ min.	Activation of Cd111 isomer
WALKER (1960)				~ 30	Absorption	Calculated from resonance parameters
WELL et al. (1950)	111	Stable	≥ 0.5	≈ 44	Absorption	Calculated from resonance parameters
WAHL (1959)	112	Stable	80 ± 40 mb	*	$\rightarrow 14$ yr	Activation of Cd113 isomer. Revised for half-life
WALKER (1960)			43 ± 10 mb	*	$\rightarrow 14$ yr	Activation of Cd113 isomer with thermal neutrons
Both the data for $\sigma_{act} \xrightarrow{14 \text{ yr}} 14 \text{ yr} \rightarrow 14 \text{ day}$ are proportional to the value assumed for $\sigma_{act} \xrightarrow{\text{Cd114}} 14 \text{ day}$, namely 0.14 ± 0.03 barns.						
MESEVER (1951)	113	Stable		2004.0	(n, γ)	Time-of-flight using pulsed cyclotron; capture radiation detection
BROOKHOUSE (1951)				19790 ± 450	Absorption	Transmission and scattering using crystal spectrometer
BNL 325 (1958) graphs				20830	Absorption	Apparently renormalisation of old transmission data by RADHATER et al. (1947)
ROSE et al (1958)			21620 ± 300	19640 ± 240	Absorption	Reactivity oscillator in thermal well; $r \approx 0$. We assume $\delta = 1.350 \pm 0.008$
WESTCOTT (1960)				X	Absorption	Tables of δ and α coefficients for natural Cd, but the absorption is due to Cd113 almost completely. The δ values may be ~ 1% low
				Recommended value 19740 ± 260		
SEVERN et al. (1947)	114	Stable	0.14 ± 0.03 1.01 ± 0.21	*	$\rightarrow 44$ day $\rightarrow 53$ hr	Activation in reactor
OVERMAN et al. (1948)			0.064 0.64	*	$\rightarrow 44$ day $\rightarrow 53$ hr	- ditto -
WALKER (1960)				≈ 10	Absorption	Calculated from resonance parameters
SEVERN et al. (1947)	116	Stable	1.3 ± 0.26	*	$\rightarrow 3.0$ hr	Activation of Cd117 isomer in reactor spectrum
SUNTAR & GOLDHABER (1949)				≈ 10	?	Probably from activation Cd-ratio of some isomer

Table 6.1 ^{49}In

References	Mass no.	Half-life	$\hat{\sigma}$ barns	σ_0 barns	Σ' barns	Reaction	Method and comments
See chapter 3	113	Stable		41 ± 19 2 ± 0.3		$\rightarrow 4.9$ day $\rightarrow 72$ sec) Activation
HARRIS et al. (1950 a)				*			Activation Cd-ratios. Recalculated
WALKER (1960)					740	$\rightarrow ?$	Absorption Calculated from resonance parameters
See chapter 3	115	6×10^{14} y		146 ± 9 50 ± 5		$\rightarrow 54$ min $\rightarrow 13$ sec) Activation
HARRIS et al. (1950 a)					2650	$\rightarrow ?$	Activation Cd-ratios. Recalculated
TATTERSALL et al. (1960)					3580 \pm 100	Absorption	Reactivity oscillators with natural In
WALKER (1960)					3230 \pm 90	Absorption	Calculated from resonance parameters
WESTCOTT (1960)							Tables of δ and s coefficients for natural In. Deviation of δ from unity is caused by the strong resonance in In ¹⁵ at 1.46 eV

Table 6.1

⁵⁰Sn

Reference	Mass no.	Half-life	$\hat{\sigma}$ barns	σ^o barns	Σ_r barns	Reaction	Method and comments
ROSE et al. (1958)	114	Stable	≤ 89	*		Absorption	Reactivity oscillator in thermal well, $r \approx 0$. Known activation cross-sections subtracted from $\sigma_{\text{natural Sn}}$
WALKER (1960)				≥ 0.15	~ 4.6	Absorption	Calculated from resonance parameters
ROSE et al. (1958)	115	Stable	≤ 170	*		Absorption	As for Sn ¹¹⁴ above
WALKER (1960)				≥ 0.04	~ 1.9	Absorption	Calculated from resonance parameters
NELSON et al. (1950)	116	Stable	6 ± 2 mb			→ 14 day	Activation in reactor of Sn ¹¹⁷ isomer
ROSE et al. (1958)			≤ 4.1	*		Absorption	As for Sn ¹¹⁴ above
WALKER (1960)				≥ 0.2	~ 13	Absorption	Calculated from resonance parameters
ROSE et al. (1958)	117	Stable	≤ 7.6	*		Absorption	As for Sn ¹¹⁴ above
WALKER (1960)				≥ 0.17	~ 15	Absorption	Calculated from resonance parameters
HILL (1951)	118	Stable	~ 3 mb	*		→ 245 day	Activation of Sn ¹¹⁹ isomer, with a reactor spectrum probably
BOWE & AXEL (1951)			11 mb	*		→ 245 day	- ditto -
ROSE et al. (1958)			≤ 2.4	*		Absorption	As for Sn ¹¹⁴ above
WALKER (1960)			> 0.09	~ 4.0	Absorption	Calculated from resonance parameters	

Table 6.1

 ^{50}Sn

continued

Reference	Mass no.	Half-life	$\bar{\sigma}$ barns	σ_0 barns	$\frac{\tau}{\tau_{1/2}}$	Reaction	Method and comments
ROSE et al. (1958)	119	Stable	≤ 6.7	*	≈ 5.6	Absorption	As for Sn^{114} above
WALKER (1960)				> 0.046		Absorption	Calculated from resonance parameters
SEREN et al. (1947)	120	Stable	55 ± 11 0.22 ± 0.04	*		$\rightarrow (400)$ day $\rightarrow 27.5$ hr	Activation in reactor
NELSON et al. (1950)			~ 1 mb	*		$\rightarrow (400)$ day $\rightarrow 27.5$ hr	- ditto -
HUGHES et al. (1953)			~ 30 mb	*		$\rightarrow 27.5$ hr	- ditto -
WALKER (1960)			0.13	*		$\rightarrow 27.5$ hr	Calculated from resonance parameters
SEREN et al. (1947)	122	Stable	0.30 ± 0.06			Absorption	
NELSON et al. (1950)			1 mb 0.1			$\rightarrow 40$ min $\rightarrow 130$ day $\rightarrow 40$ min	Activation in reactor Activation
HUGHES et al. (1953)			0.16 ± 0.04			$\rightarrow 40$ min ?	Activation in reactor
SUNTAR & GOLDHABER (1949)					$\lesssim 10$ Small	Absorption	Probably from activation Cd-ratios No resonance known below 100 eV
SEREN et al. (1947)	124	Stable	0.15 ± 0.03 0.65 ± 0.13			$\rightarrow 9.7$ day $\rightarrow 9.6$ min	Activation in reactor
NELSON et al. (1950)			2 mb			$\rightarrow 9.7$ day $\rightarrow 9.6$ min	Activation
HUGHES et al. (1953)			0.5 mb			$\rightarrow 9.7$ day $\rightarrow 9.6$ min	Activation in reactor
WALKER (1960)			5 mb 0.15			$\rightarrow 9.7$ day $\rightarrow 9.6$ min	Calculated from resonance parameters
			> 0.18	~ 12	Absorption		

Upper limits are given above for the thermal cross-sections of Sn^{114} to Sn^{116} . These data are correlated, so that if the maximum value holds for one isotope the other two must have very small cross-sections.

Table 6.1

 Sb

Reference	Mass no.	Half-life	δ barns	σ_0 barns	$\Sigma' r$ barns	Reaction	Method and comments
SEREN et al. (1947)	121	Stable	6.8 ± 1.4	*		$\rightarrow 2.8$ day	Activation in reactor. Revised to allow for 3.1% decay of Sb122 by orbital electron capture.
POMERANCE (1952)				5.9 ± 0.5		Absorption	Local oscillator; $r \approx 0$
SEGAL et al. (1959)			0.19 ± 0.03	*		$\rightarrow 3.5$ min	Activation in thermal column
WALKER (1960)				$\gtrsim 5.3$	≈ 200	Absorption	Calculated from resonance parameters
The 3.5 min isomeric state of Sb ¹²² decays by isomeric transition to the ground state. The value given above for δ_{act} [$\rightarrow 2.8$ day] is for production of the 2.8 day ground state both directly and by decay of the short-lived isomer.							
SEREN et al. (1947)	123	Stable	2.6 ± 0.5	*		$\rightarrow 61$ day	Activation in reactor. Includes decay of isomeric states
der MATFOSIAN et al. (1947)			~ 30 mb	*		$\rightarrow 1.3$ min	Activation with thermal neutrons
OVERMAN et al. (1948)			4.4	*		$\rightarrow 61$ day	Activation in reactor
POMERANCE (1952)				4.1 ± 0.3		Absorption	Local oscillator; $r \approx 0$
ANL unpublished work quoted in BNL 325 (1955)			30 ± 15 mb	*		$\rightarrow 21$ min	Activation
WALKER (1960)				$\gtrsim 2.0$	≈ 160	Absortion	Calculated from resonance parameters
MURIN et al. (1957)	124	61 day	~ 500			$\rightarrow 2.0$ yr	Activation in reactor. Detection of 58 day Tl ^{125m} daughter activity. Revised for product half-life.

Table 6:¹⁸52¹⁸

Reference	Mass no.	Half-life	δ , barns	σ_0 , barns	Σ'_{r} , barns	Reaction	Method and comments
HILL (1949)	122	Stable	1.2 ± 0.5	*		$\Rightarrow 112$ day	Activation in reactor of Te ¹²³ isomer. Revised for half-life of product
POMERANCE (1952)						Absorption	Local oscillator; $r \approx 0$
WALKER (1960)						Absorption	Calculated from resonance parameters
POMERANCE (1952)	123	Stable		4.04 ± 3.2		Absorption	Local oscillator; $r \approx 0$
WALKER (1960)				$\gtrsim 470$	≈ 6560	Absorption	Calculated from resonance parameters Strong resonance at 2.33 eV
HILL (1949)	124	Stable	5 ± 3	*		$\Rightarrow 58$ day	Activation in reactor of Te ¹²⁵ isomer
POMERANCE (1952)				6.7 ± 1.3		Absorption	Local oscillator; $r \approx 0$
WALKER (1960)					~ 2.1	Absorption	Calculated from resonance parameters
POMERANCE (1952)	125	Stable		1.54 ± 0.15		Absorption	Local oscillator; $r \approx 0$
WALKER (1960)				*	~ 2.3	Absorption	Calculated from resonance parameters
SEREN et al. (1947)	126	Stable	75 ± 15 mb	*		$\Rightarrow 105$ day	Activation in reactor
POMERANCE (1952)			0.80 ± 0.16	*		$\Rightarrow 9.4$ hr	
WALKER (1960)				0.80 ± 0.20		Absorption	Local oscillator; $r \approx 0$
SEREN et al. (1947)					~ 9	Absorption	Calculated from resonance parameters
POMERANCE (1952)	128	Stable	16 ± 3 mb	*		$\Rightarrow 34$ day	Activation in reactor
WALKER (1960)			0.14 ± 0.03	*	0.3 ± 0.3	$\Rightarrow 72$ min	Local oscillator; $r \approx 0$
SEREN et al. (1947)				$\gtrsim 0.9$	~ 10	Absorption	Calculated from resonance parameters
POMERANCE (1952)	130	Stable	< 9 mb	*		$\Rightarrow 30$ hr	Activation in reactor
WALKER (1960)			0.21 ± 0.04	*	0.5 ± 0.3	$\Rightarrow 25$ min	Local oscillator; $r \approx 0$
SEREN et al. (1947)						Absorption	
POMERANCE (1952)							

Table 6.1

53^I

Reference	Mass no.	Half-life	σ barns	σ_0 barns	Σ'_{r} barns	Reaction	Method and comments
SEREN et al. (1947)	127	Stable	6.65 ± 1.3	*		\rightarrow 25 min	Activation in thermal column. Corrected for 6.4% decay of T^{128} by orbital electron capture
WU et al. (1947)				7.04 ± 0.31		Absorption	Transmission, using pulsed cyclotron
JONES (1947)				6.64 ± 0.25		Absorption	- ditto -
POMERANCE (1951)				6.3 ± 0.32		Absorption	Local oscillator; $r \approx 0$
GRIMELAND (1952)				5.72 ± 0.3	*	\leftrightarrow 25 min	Activation in thermal column, relative to $\sigma_{\text{act}} [\text{Na}] = 0.534$ barns. $\delta = 1.000$ assumed
TATTERSALL et al. (1960)				5.72 ± 0.3		Absorption	Reactivity oscillator in thermal well; $r \approx 0$. $\delta = 1.00$ assumed
HARRIS et al. (1950 a)				6.6 ± 0.3	*		Activation Cd - ratio. Recalculated Reactivity oscillator in thermal well and lattice spectra
TATTERSALL et al. (1960)							Calculated from resonance parameters
WALKER (1960)							
					130	\rightarrow 25 min	
					180 ± 30	Absorption	
					160 ± 30	Absorption	
					160 ± 30		
PURKAYASTHA & MARTIN (1956)	129	1.7×10^7 yr	36	*		\rightarrow 12.5 hr	Activation in reactor. Revised using $\sigma_{\text{act}} [\text{Cu}^{63} \rightarrow 12.8 \text{ nr } \beta] = 2.55 \pm 0.06$ barns
ROY & WUSCHKE (1958)				27.8 ± 2.1			\rightarrow 12.5 hr
BLOCK et al. (1960)				27.8 ± 2.1		Absorption	Activation with sub-od neutrons. Revised $\delta = 1.000$ assumed
PATTERDEN (1960)				31 ± 5		Absorption	Transmission using a fast chopper
ROY & WUSCHKE (1958)				28.0 ± 2		Absorption	- ditto -
WALKER (1960)					25 ± 5.8	\rightarrow 12.5 hr	Activation Cd-ratio, with revised δ Calculated from resonance parameters
					28 ± 7.6	Absorption	
ROY & WUSCHKE (1958)	130	12.5 hr	18.9 ± 3.1			\rightarrow 8.1 day	Activation in reactor; $r_b = 0.032$. Revised
KENNEDY & THOMS (1958)	131	8.1 day	54 ± 42			(n, γ)	Mass spectrometry. Revised

Table 6.1

54

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ^*_R barns	Reaction	Method and comments
MACHAMARA & THODE (1950)	128	Stable	< 5		Small?	(n, γ) Absorption	Mass spectrometry. Reactor spectrum No resonance known
MACHAMARA & THODE (1950) WALKER (1960)	129	Stable	4.5 ± 15		~ 240	(n, γ) Absorption	Mass spectrometry. Reactor spectrum Calculated from resonance parameters
MACHAMARA & THODE (1950)	130	Stable	< 5		Small?	(n, γ) Absorption	Mass spectrometry. Reactor spectrum No resonance known
MACHAMARA & THODE (1950) PALMER (1960) NEPHEW (1960) }	131	Stable	120 ± 15		760 ± 175	(n, γ) Absorption	Mass spectrometry. Reactor spectrum Calculated from resonance parameters
KERN & SNELL (1943)	132	Stable	0.2 ± 0.1			5.27 day	Activation. Probably includes production through decay of the 2.3 day Xe^{133} isomer
MACHAMARA & THODE (1950)			< 5		Small?	(n, γ) Absorption	Mass spectrometry. Reactor spectrum No resonance known
KENNEDY & THODE (1958)	133	5.27 day	198 ± 94			(n, γ)	Mass spectrometry. Reactor spectrum. Revised
TURKEVICH et al. (1944)	134	Stable	0.2 ± 0.1			$\rightarrow 9.18$ hr	Activation. Probably includes production through decay of the 15 min Xe^{135} isomer
MACHAMARA & THODE (1950)			< 5		Small?	(n, γ)	Mass spectrometry. Reactor spectrum No resonance known
TURKEVICH et al. (1945)	135	9.18 hr		See section 6.5			Activation in reactor spectrum. 30 yr Cs^{137} daughter activity measured. Result corrected for Cs^{137} half-life and for conversion electrons from Ba^{137m}
					Small?	Absorption	No resonance known

Table 6.1

⁵⁵Cs

Reference	Mass no.	Half-life	σ barns	σ_0 barns	Σ_r^* barns	Reaction	Method and comments
SEREN et al. (1947)	133	Stable	16 ± 3 mb	*	*	$\rightarrow 3.2$ hr $\rightarrow 2.17$ hr	Activation in reactor - Ditto - Revised half life
HARRIS et al. (1950 b)			32 ± 6	*		Absorption	Reactivity oscillator in reactor spectrum; $r_0 = 0.046$
POLTERANCE (1951)			38.2	*	30.0 ± 1.5	Absorption	Local oscillator; $r \approx 0$
ROSE et al. (1958)			28 ± 1	*		Absorption	Reactivity oscillator in thermal well; $r \approx 0$, $\delta = 1.002$ at 20°C
STUDNOSTI et al. (1958)			41.7	*		(n, γ)	Mass spectrometry. Reactor spectrum with $r_0 = 0.03$. Revised
BATTY et al. (1958)			31.8 ± 1.2	*		$\rightarrow 2.17$ yr	Activation in reactor; $r_0 \approx 0.02$. Revised
Chalk River unpublished work quoted in BILL 325 (1960)			3.0 ± 0.3	*		$\rightarrow 3.2$ hr	Activation of Cs^{134} m isomer
KULENTOV & GRYAZEV (1957)					195 ± 35	Absorption	Reactivity measurements sub-Cd. Revised
TATTERSALL et al. (1960)					490 ± 80	Absorption	Reactivity oscillator in thermal well and lattice spectra
NEWPHAM (1960)					375 ± 75	Absorption	Calculated from resonance parameters
BATTY et al. (1958)	134	2.17 yr	193 ± 17		29 ± 1	(n, γ)	Mass spectrometry; $f_{33} \approx 0.03$. Revised using δ [Cs^{133}] = 41.7 barns
SUGARMAN (1949)	135	2.6×10^5 y	14.5 ± 4.4			$\rightarrow 13$ day	Activation
BAERG et al. (1958)			9.1 ± 0.5			$\rightarrow 13$ day	Mass spectrometry and activations in sub-Cd spectrum. Revised
- ditto -					56.3 ± 2.4	$\rightarrow 13$ day	- Ditto - , with Cd-ratios
COCKING (1958)					Small	Absorption	Transmission using fast chopper No resonance found below 52 ev
STUPEGIA (1960)	137	30 yr	0.11 ± 0.05			$\rightarrow 32$ min	Activation in reactor tamper; $r_0 = 0.0025$

Table 6.1

Reference	Mass no.	Half-life	σ barns	σ_{∞} barns	τ^{ir} barns	Reaction	Method and comments
POMERANCE (1952)	134	Stable		2 ± 2		Absorption	Local oscillator; $r \approx 0$
POMERANCE (1952)	135	Stable		5.8 ± 0.9		Absorption	As for Ba^{134}
POMERANCE (1952)	136	Stable		0.4 ± 0.4		Absorption	As for Ba^{134}
WALKER (1960)			*	*	≈ 13	Absorption	Calculated from resonance parameters
POMERANCE (1952)	137	Stable		5.1 ± 0.4		Absorption	As for Ba^{134}
SEREN et al. (1947)	138	Stable	0.51 ± 0.10	*	Small	No resonance known below 100 eV	
POMERANCE (1952)				0.70 ± 0.11		$\rightarrow 24$ min	Activation in thermal column
SUNTAR & GOLDHABER (1949)					$\lesssim 10$	Absorption	As for Ba^{134}
KATCOFF (1945)	139	84 min	3.8 ± 1		Small	?	Probably from activation cadmium ratio
YAFFE et al. (1949)			4.7 ± 1.2			$\rightarrow 12.8$ day	Activation
EASTWOOD et al. (1958)	140	12.8 day	<20			$\rightarrow 12.8$ day	Activation in reactor
						$\rightarrow 18$ min	Activation in reactor; $r \approx 0.02$. 32 day C_{141} activity was measured

Table 6.1

57

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ barns	Reaction	Method and comments
BOTHE (1946)	139	Stable	> 8.5	*		\rightarrow 40 hr	Activation
SEREN et al. (1947)			8.4 ± 1.7	*		\rightarrow 40 hr	Activation in reactor
WATTENBERG & THOMAS (1948)			7.6	*			Reactivity measurements in reactor spectrum. Revised using $\sigma_A [B] = 758$ barns
HARRIS et al. (1950 b)			9.62	*			Reactivity oscillator; $r_o = 0.047$
POTTERANCE (1951)			9.1 ± 0.6				Local oscillator; $r \approx 0$
BENOIST et al. (1951)			8.93 ± 0.11	*			Reactivity oscillator in reactor spectrum. $r_o = 0.031$
ROSE et al. (1958)			8.9 ± 0.3	8.9 ± 0.3	25 ± 8		Reactivity oscillator in thermal well; $r \approx 0$. $g = 1.00$ assumed
NEPHEW (1960)							Calculated from resonance parameters
KATCOFF et al. (1949)	140	40 hr	3.1 ± 1.0	9.0 ± 0.3		3.8 hr.	Activation in reactor temper

58^{Ge}

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ' barns	Reaction	Method and comments
BOTHE (1946)	140	Stable	> 0.55	*		\rightarrow 33 day	Activation
KATCOFF et al. (1949)			0.31 ± 0.07	*		\rightarrow 33 day	Activation in reactor tamper
HUGHES & SHERMAN (1950)			0.24	*		\rightarrow 33 day	Activation in reactor spectrum
POMERANCE (1952)				0.65 ± 0.07		Absorption	Local oscillator; $r \approx 0$. Samples were contaminated with Cd; see ROY & YAFFE (1956 b)
KLIMENTOV & GRYAZEV (1957)					$\lesssim 5$	Absorption	Reactivity measurements with natural Ce under Cd. Revised
BIDINOSTI et al. (1958)	141	33 day	A few hundred			(n, γ)	Mass spectrometry. Reactor spectrum
BOTHE (1947)	142	5×10^{15} yr	1.3	*		\rightarrow 33 hr	Activation
KATCOFF et al. (1949)			0.95 ± 0.18	*		\rightarrow 33 hr	Activation in reactor tamper
HUGHES & SHERMAN (1950)			0.72	*		\rightarrow 33 hr	Activation in reactor
POMERANCE (1952)				1.82 ± 0.27		Absorption	As for ^{140}Ce , above
ROY & YAFFE (1956 b)			0.29 ± 0.05	*		\rightarrow 33 hr	Activation in reactor where spectrum was predominantly thermal
ORNL unpublished work quoted in BNL 325 (1960)			0.94 ± 0.05	*		\rightarrow 33 hr	Activation
KLIMENTOV & GRYAZEV (1957)					$\lesssim 4.0$	Absorption	As for ^{140}Ce above
ROY & YAFFE (1956 a)	143	33 hr	6.3 ± 0.7			\rightarrow 280 day	Activation in reactor. Revised

A somewhat lower value was obtained from activation in a low neutron flux. The difference may have been due to experimental uncertainties. On the other hand the higher values may have resulted from resonance absorption in ^{143}Ce . This alternative interpretation leads to the values: $\delta = 4.4$ barns for thermal neutrons, $\Sigma' \lesssim 16$ barns

Table 6.1

⁵⁹Pt

Reference	Mass no.	Half-life	δ barns	σ_0 barns	E_r' barns	Reaction	Method and comments
BOTHE (1946)	141	Stable	16	*		$\rightarrow 19.2$ hr	Activation
SEREN et al. (1947)			10.1 ± 2	*		$\rightarrow 19.2$ hr	Activation in thermal column
POMERANCE (1951)				<u>11.6 ± 0.6</u>			Local oscillator; $r \approx 0$
ROSE et al. (1958)			11.5 ± 0.3	*			Reactivity oscillator in thermal well; $r \approx 0$
ORNL unpublished work quoted in BNL 325 (1960)			10.8 ± 1.0	*	<u>6.6</u> ≈ 19	$\rightarrow 19.2$ hr	Activation
HARRIS et al. (1950 a)						$\rightarrow 19.2$ hr	Activation Cd ratio. Recalculated
WALKER (1960)						Absorption	from resonance parameters
SMITH & REEDER (1955)	142	19.2 hr	20 ± 3			$\rightarrow 13.7$ day	Activation. $s = 0$ assumed. Revised
BIDINOSTI et al. (1958)	143	13.7 day	400 ± 200		(n, γ)		Mass spectrometry. Reactor spectrum
ROY & ROY (1959)			91 ± 10		$\rightarrow 17.3$ min	Activation with sub-Cd spectrum	
- ditto -					$\rightarrow 17.3$ min	Activation Cd-ratios	
				178 ± 25			

Table 6.1

60th

Table 6.1

 ^{60}Ni continued

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ_r' barns	Reaction	Method and comments
MARINSKY & GLENDENIN (1945)	146	stable	1.0 ± 0.3	*		$\rightarrow 11.1$ day	Activation in reactor
BOTHE (1946)			1.7 ± 0.4	*		$\rightarrow 11.1$ day	Activation
POMERANCE (1952)				10.2 ± 0.8		Absorption (n, γ)	Local oscillator; $r \approx 0$
WALKER & THODE (1953)			3.5 ± 5	*			Mass spectrometry. Reactor spectrum
MARINSKY & GLENDENIN (1945)	148	stable	3.2 ± 1.0	*		$\rightarrow 1.8$ hr	Activation in reactor
BOTHE (1946)			3.9 ± 1.0	*		$\rightarrow 1.8$ hr	Activation. Both product and daughter activities observed
POMERANCE (1952)				3.4 ± 1.0		Absorption (n, γ)	Local oscillator; $r \approx 0$
POMERANCE (1952)				*			Mass spectrometry. Reactor spectrum
WALKER & THODE (1953)			3.5 ± 9			Absorption	No resonance known below 80 eV
MARINSKY & GLENDENIN (1945)	150	stable	1.3 ± 0.4	*		$\rightarrow 14$ min	Activation in reactor
POMERANCE (1952)				3.0 ± 1.5		Absorption (n, γ)	Local oscillator; $r \approx 0$
WALKER & THODE (1953)				*			Mass spectrometry. Reactor spectrum
SENGAL et al. (1959)				1.5 ± 0.2		$\rightarrow 14$ min	Activation in thermal column
					Small	Absorption	No resonance known below 80 eV

Table 6.1

61^{Pr}

continued

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ_r' barns	Reaction	Method and comments
PARKER et al. (1947)	147	2.65 yr	60 ± 20			$\rightarrow 5.3$ day (n, γ)	Activation in reactor
ANIKINA et al. (1958)			90 ± 20				Mass spectrometry. Reactor spectrum
BIDINOSTI et al. (1928)			209 ± 50				- ditto - Revised
HARVEY et al.							Transmission using neutron chopper. Tentative result
WALKER & NEPHEW (1960)				180			
				≥ 78	2050 ± 400) Calculated from resonance parameters
						Recommended value	140 ± 40

The activation result is low if Pm¹⁴⁸ decays appreciably to Nd¹⁴⁸ by orbital electron capture.

Table 6.1

62 Sm

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ_r' barns	Reaction	Method and comment
GORSHEKOV et al. (1957)	147	1.3×10^{11} yr	1000 ± 50			(n, γ)	Mass spectrometry of fission-products produced in a reactor. This result was omitted from a subsequent report by ANIKINA et al. (1958)
JOWITT et al. (1958)			87 ± 60	*		Absorption	Reactivity oscillator in thermal well; $T \approx 0$
PATTERDEN (1960)				150 ± 210		Absorption	Transmission using neutron chopper. Sm150 contribution subtracted
TATTERSALL et al. (1960)					<1350	Absorption	Reactivity oscillator in thermal well and lattice spectra
WALKER (1960)					680	Absorption	Calculated from resonance parameters
MOREYARDS & ANDERSEN (1954)	149	Stable		39200		Absorption	Transmission with natural Sm, using a crystal spectrometer; from graphical data in BNL 325 (1958). Small contribution from other isotopes subtracted
PATTERDEN (1958)				39900 ± 600		Absorption	Transmission, using a crystal spectrometer
ROSE et al. (1958)			68200 ± 700	41200 ± 1000		Absorption	Reactivity oscillator in thermal well; $r \approx 0$. $T = 20^\circ\text{C}$, $\epsilon = 1.65 \pm 0.036$ assumed
PATTERDEN (1958) WESTCOTT (1960)				Recommended value 39800 ± 900		Absorption	Tables of ϵ and δ coefficients
							Because of this δ in a reactor depends strongly on the effective temperature of the thermal neutron component. This sensitivity has been exploited by TATTERSALL et al. (1960). Measurements of δ in reactor spectra have been reported also by BIDINOSTI et al. (1958) and by others, but are better regarded as spectral indicators, so the results are not listed here.

Table 6.1

62 Sm

continued

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ_r barns	Reaction	Method and comments
CORNISH & ATKEN (1958)	150	Stable	102 ± 10	*		(n, γ)	Mass spectrometry. Reactor spectrum
PATTERDEN (1960)				200 ± 220			Transmission using neutron chopper. Sm 147 contribution subtracted
INGHRAM et al. (1950)	151	~ 90 yr	12400		Small	(n, γ)	No resonance known below 21 eV
MELAIKA et al. (1955)			12000			(n, γ)	Mass spectrometry. Reactor spectrum
ANIKINA et al. (1958)			8000 ± 1500			(n, γ)	Mass spectrometry of fission-products produced in a reactor
BIDINOSTI et al. (1958)			12600			(n, γ)	Mass spectrometry. Reactor spectrum
COCKING (1958)				11500 ± 1500	Absorption	Transmission with neutron chopper	
PATTERDEN (1960)				15400 ± 1500	Absorption	Transmission with neutron chopper Two samples	
			Recommended value	14800 ± 1300	Absorption	Not 1/v in thermal region	
BOTHE (1946)	152	Stable	~ 340			→ 47 hr	Activation
SEREN et al. (1947)			134 ± 27			→ 47 hr	Activation in thermal column
PATTERDEN (1958)				200 ± 6	Absorption		Transmission with crystal spectrometer
ROSE et al. (1958)			224 ± 7	231 ± 7	Absorption		Reactivity oscillator in thermal well; $r \approx 0$. $\epsilon = 0.970$ at 20°C
TATTERSALL et al. (1960)				2850 ± 300	Absorption		Reactivity oscillator in thermal well and lattice spectra. Mainly due to 8.1 eV resonance
HARRIS et al. (1950a)					2410	→ 47 hr	Activation Cd-ratio. Revised
WALKER (1960)					2100	Absorption	Calculated from resonance parameters
			Recommended value	213 ± 14			
SEREN et al. (1947)	154	Stable	4.9 ± 1.0	*	Small	→ 24 min.	Activation in thermal column
						Absorption	No resonance known below ~ 100 eV

Table 6.1

 ^{65}Eu

Reference	Mass no.	Half-life	δ barns	Σ_{f} barns	Reaction	Method and comments
BOTYEE (1946)	151	Stable	2500 ± 600		$\rightarrow 9.2 \text{ hr}$	Activation. Corrected to allow for 23% decay of 9.2 hr Eu^{152} by orbital electron capture, and for revised standard.
SERENI et al. (1947)			1850 ± 370		$\rightarrow 9.2 \text{ hr}$	Activation in thermal column. Corrected for electron capture decay of product.
SERENI et al. (1947)			4400 ± 1100		$\rightarrow 12.6 \text{ yr}$	Activation in thermal column. Corrected for Eu^{154} activity and for 73% decay of 13 yr Eu^{152} by electron capture.
OVERMAN et al. (1948)			6830		$\rightarrow 12.6 \text{ yr}$	Activation in reactor. Corrected as above.
LADDEN et al. (1949)			1700 ± 260		$\rightarrow 9.2 \text{ hr}$	Mass spectrometer and activation studies. Renormalised to give δ [natural] $\text{Eu} = 4470 \text{ b}$ in reactor spectrum.
STURM (1947)			7200 ± 1100		$\rightarrow 12.6 \text{ yr}$	Absorption
PATTERDEN (1958)			9000			Transmission with natural Eu using crystal spectrometer; from graphical data in BNL 325 (1958). Eu^{153} contribution subtracted.
TATTERSALL et al. (1960)				7700 ± 80	Absorption	Transmission using crystal spectrometer
TATTERSALL et al. (1960)				8810 ± 90	Absorption	Reactivity oscillator in thermal well; $r = 0$. $\delta = 0.9981$ assumed
TATTERSALL et al. (1960)				3000	Absorption	Reactivity oscillator in thermal well and lattice spectra
TATTERSALL et al. (1960)				3470 ± 300	Absorption	Calculated from resonance parameters
				8160 ± 350	Recommended values	

Table 6.1

63 Eu

continued

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Z_r barns	Reaction	Method and comments
HAYDEN et al. (1949)	152	12.6 yr	5500 ± 1500			(n, γ)	Mass spectrometry. Reactor spectrum. Renormalised as for Eu ¹⁵¹ above
GLENDEEN & MARINSKY (1949)	153	Stable	410 ± 100 630			(n, γ) - 16 yr	As for Eu ¹⁵¹ above Activation in reactor. Revised for half-lives involved
PATTERDEN (1958)				448 ± 16	X	Absorption	Transmission with crystal spectrometer
PATTERDEN (1958)						Absorption	Tables of ε and s coefficients
TATTERSALL et al. (1960)				309 ± 5		Absorption	Reactivity oscillator in thermal well; $r \approx 0$. $\varepsilon = 1.026$ at 20°C assumed
PATTERDEN (1958)					1780 1400 ± 300 1280 ± 100	Absorption Absorption Absorption	Calculated from resonance parameters
WALKER (1960)							Reactivity oscillator in thermal well and lattice spectra
TATTERSALL et al. (1960)					400 ± 50	1400 ± 300	Absorption
			Recommended values				
HAYDEN et al. (1949)	154	16 yr	1510 ± 300			(n, γ)	As for Eu ¹⁵¹ above
INGRAM et al. (1947)	155	1.7 yr	~ 14000			(n, γ)	Mass spectrometry. Reactor spectrum
MARINSKY (1949)			~ 30000				From expected and observed fission-product activities
HAYDEN et al. (1949)			13600 ± 3000			(n, γ)	As for Eu ¹⁵¹ above

Table 6.1

Gd

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ_r' barns	Reactions	Method and comments
ROSE et al. (1958)	155	Stable	49800 \pm 600	58300 \pm 700		Absorption	Reactivity oscillator in thermal well; $r \approx 0$. $\epsilon = 0.854$ at 20°C assumed
PATTENDEN (1958)				66000 \pm 2000	XX	Absorption	Transmission using crystal spectrometer
PATTENDEN (1958)			Recommended value	62000 \pm 3500		Absorption	Tables of ϵ and s coefficients
WALKER (1956)	156	Stable	< 14	*	Small	Absorption	Reactor neutrons No resonance known below 3.7 eV
BRILL & LICHTENBERGER (1947)	157			233000 \pm 7000		Absorption	Transmission with natural Gd using neutron chopper; from graphical data in BNL 325 (1958) isotopes subtracted
ROSE et al. (1958)		Stable		213000 \pm 2000	249000 \pm 2000	Absorption	Reactivity oscillator in thermal well; $r \approx 0$. $\epsilon = 0.854$ at 20°C assumed
PATTENDEN (1958)				264000 \pm 4500	XX	Absorption	Transmission using crystal spectrometer
PATTENDEN (1958)			Recommended value	255000 \pm 8500		Absorption	Tables of ϵ and s coefficients
SEREN et al. (1947)	158	Stable		3.6	*	\rightarrow 18.0 hr	Activation in reactor
BUTEMENT (1949)				4.5		\rightarrow 18.0 hr	Activation in reactor
ORNL unpublished work quoted in Brill 325 (1960)				3.9 \pm 0.4		\rightarrow 18.0 hr	Activation
SEREN et al. (1947)	160	Stable	\sim 2.7		Small	Absorption	No resonances known below 3.7 eV
BUTEMENT (1949)			0.8			\rightarrow 3.6 min	Activation of Tb^{161} daughter in reactor
						\rightarrow 3.6 min	Activation in reactor. Both product and daughter activities observed

LAPP et al. (1947) first showed, by mass spectrometric analyses, that the neutron absorption in natural Gd is due mainly to the isotopes Gd¹⁵⁵ and Gd¹⁵⁷. INGERMAN et al. (1950) attempted to derive the cross-sections of these two isotopes from a mass spectrometric study of fission products. However the derivations are extremely complex and rely on assumed fission yield values. We have ignored these results which add nothing to the data listed.

Table 6.1

^{65}Tb

Reference	mass no.	Half-life	δ barns	σ_0 barns	Σ_f barns	Reaction	Method and comments
BOTT (1945)	159	Stable	> 27	*		\rightarrow 73 day Absorption	Activation Local oscillator; $r \approx 0$
POURANCE (1951)				46 ± 5	370		Calculated from resonance parameters
WALKER (1960)							
SUTTE et al. (1956)	160	73 day	575 ± 100			\rightarrow 6.9 day	Activation in reactor spectrum

Table 6.1

 ^{66}Dy

Reference	Mass no.	Half-life	δ barns	σ_0 barns	Σ' barns	Reaction	Method and comments
WALKER (1956)	160	Stable	100 \pm 70	*	Small		Reactor spectrum No resonance known below 40 eV
WALKER (1956)	161	Stable	620 \pm 35	\gtrsim 40	950		Reactor spectrum Calculated from resonance parameters
WALKER (1956)	162	Stable	330 \pm 90	*	210	2570	Reactor spectrum Calculated from resonance parameters
WALKER (1956)	163	Stable	420 \pm 150	\gtrsim 90	1230		Reactor spectrum Calculated from resonance parameters
ATTERLING et al. (1945)	164	Stable	2460	*		\rightarrow 139 min	Activation
BOTHE (1946)			2680 \pm 300	*		\rightarrow 139 min	Activation
FLAMMERSFIELD (1946)			2150	*		\rightarrow 1.25 min	Activation
SEREN et al. (1947)			2580 \pm 260	*		\rightarrow 139 min	
BRILL & LICHTENBERGER (1947)				<u>2690 \pm 200</u>			Absorption
POTTERANCE (1951)					\approx 2290		Absorption $\text{Dy}; r \approx 0$ Contributions of other isotopes subtracted
WEBER (1954)			1860 \pm 100	*		\rightarrow 1.25 min	Activation
SEHGAL et al. (1959)			2520 \pm 300	*		\rightarrow 1.25 min	Activation with sub-Cd neutrons.
			< 3360			\rightarrow 139 min) Revised using $\delta[\text{Au}] = 98.9$ barns
							The 1.25 min isomeric state of ^{165}Dy decays mainly by isomeric transition to the 139 min ground state. The values given above for δ [\rightarrow 139 min] are for production of the ground state both directly and by decay of the short-lived isomer.
KETELLE (1949)	165	139 min	5000			\rightarrow 81 hr	Activation in reactor
BUTEMENT (1950)			1210			\rightarrow 81 hr	Activation in reactor. Revised

The slow neutron absorption cross-section $\sigma_A(E)$ may be calculated from the resonance parameters. The cross-section below ~ 1 eV is a perfect example of an isolated resonance, to which the Breit-Wigner single-level formulae may be applied with confidence, and with the simplification that the potential scattering is quite negligible. Unfortunately the spin of the resonant state is not known. The spin of the Xe^{135} ground-state is probably $3/2$, THULIN (1955), so the statistical weight factor g^* is either $3/8$ or $5/8$. With either value of g^* , resonance parameters may be evaluated which give almost equally good fits to the total cross-section data. Thus to each of the two studies of $\sigma_T(E)$ correspond two sets of resonance parameters; hence there are four possible curves for $\sigma_A(E)$. The four values for σ_0 are given in Table 6.2.

No direct measurement of the scattering or absorption cross-section of Xe^{135} has been reported. However a considerable number of measurements of σ_A have been made in broad neutron spectra in reactors:

- (i) By observing reactivity transients either at start-up or in decay. This sort of measurement determines $y \sigma_A$ where y is the yield of Xe^{135} per fission.
- (ii) By "de-activation" methods; measurements of the loss of Xe^{135} activity caused by neutron capture in a high flux irradiation.
- (iii) By mass-spectrometer measurements. Usually the long-lived daughter Cs^{135} has been observed.

The most recent measurements are listed in Table 6.3. The best data appear to be those of FICKEL & TOMLINSON (1959). In a reactor $\sigma_A [Xe^{135}]$ must be fairly sensitive to both the shape of the thermal component of the neutron spectrum and to that of the slowing-down component below about 0.3 eV. Therefore the data collected in table 6.3 must be used with caution. The better measurements appear to support the data of SMITH et al. (1959) with $g^* = 5/8$, but not quite decisively. If $g^* = 3/8$ the data of BERNSTEIN et al. (1956) seem to be ruled out.

Besides the data listed in table 6.3 several earlier measurements have been cited, by FREEDMAN et al. (1945), and by BERNSTEIN & SMITH (1955). Probably many of the results quoted were given in the "true flux" convention rather than in that of WESTCOTT (1958). However details of the flux calibrations are not available and the accuracy claimed was not very high so we have ignored these early data.

Table 6.2, ${}^{54}\text{Xe}^{135}$

Absorption cross-section at 2200 eV/sec from transmission measurements

Reference	σ_A at 2200 m/sec 10^6 barns	$g^* = 3/8$	$g^* = 5/8$	Method and comments
BERNSTEIN et al. (1956)	2.19 ± 0.10	2.40 ± 0.10		Transmission, using bent crystal focussing spectrometer. Samples of fission-product iodine were separated in the form Pd I ₂ , with Xe ₁₃₅ as a decay product. Assay by absolute β counting and comparative γ counting of I activities, and also by absolute $\gamma\beta$ coincidence counting of Xe ₁₃₅ ; there was good agreement. Resonance analysis not corrected for Doppler effect
SMITH et al. (1959)	2.50 ± 0.06	2.76 ± 0.06		Transmission, using fast chopper. Samples were of fission-product Xe. Assay was by mass-spectrometry of the Cs ¹³⁵ daughter, with isotopic dilution, after decay of all the Xe ₁₃₅ . Resonance analysis was corrected for Doppler effect

- (i) The recommended expedient is to use the data of SMITH et al. (1959) with either $g^* = 3/8$ or $5/8$. The data of table 6.2 tend to support the higher value of g^* , but not decisively.
- (ii) The symbol g^* used above is the spin statistical factor in the Breit-Wigner formulae. Tables of the g -coefficient for a Maxwellian spectrum are given by WESTCOTT (1958 and 1960).
- (iii) The resonance absorption integral must depend sensitively on the shape of the slowing-down spectrum below 0.3 eV. As a rough guide $\Sigma' / \sigma_0 \sim 0.3$; Σ' includes the $1/v$ component.

Table 6.3. $^{54}\text{Xe}^{135}$

Absorption cross-section in various thermal reactor spectra

Reference	δ_A 10^6 barns	Spectrum $T_{^0\text{C}}$ and r	Weight	Method and comments
FREEDMAN et al. (1945)	3.02 ± 0.24	130°C sub-Cd $r \sim 0.03$	0.1	De-activation by irradiation in CP3 reactor. Result virtually for sub-Cd spectrum as the sub-Cd flux was compared with that in calibrated standard pile by Au activations. Flux also estimated from Ba-140 activity induced in natural U monitor, but cross-section assumed is not reported. Result amended to WESTCOTT convention
PETRUSKA et al. (1955)	3.21 ± 0.15	$(138 \pm 10)^\circ\text{C}$ $r \sim 0.01$	0.2	Mass-spectrometer comparison of Cs-135/Cs-137 abundance ratio in fission products formed at high and low fluxes. Original results revised by DEUTSCH (1956) to give $\delta_A[\text{Xe}^{135}] / \delta_A[\text{Ba}^{10}] = 834 \pm 40$. The spectrum given here follows the comments of FICKEL & TOMLINSON (1959); see below
IVANOV et al. (1957)	3.2 ± 1.0		0	Mass-spectrometric measurements of Cs-135 yield in highly irradiated sample of U-233. Details of flux calibration and neutron spectrum are not available
HOPKINS & JAMIESON (1957 and 1958)	2.34 ± 0.12	$\sim 20^\circ\text{C}$	0	Reactivity measurements in low power reactor gave $y[\text{I}^{135}] \cdot \delta_A[\text{Xe}^{135}] = 1.48 \times 10^5$ barns. Very complex flux determination appears to ignore cumulative effect of stable fission-product poisons in highly irradiated U monitor. We assume $y[\text{I}^{135}] = 6.32\%$

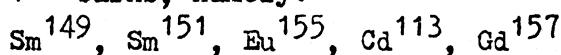
Table 6.3, $^{54}\text{XE}^{135}$, continued

Reference	$10^6 \delta_A$	Spectrum $T^\circ\text{C}$ and r	Weight	Method and comments
MAKAROV & SAMOLOVA (1958)	1.75 ± 1.15	$530 \pm 50^\circ\text{C}$	0	De-activation measurement. Details not reported. Cross-section given might be in the "true flux" convention
FICKEL & TOMLINSON (1959)	3.26 ± 0.06 3.39 ± 0.11	$\begin{cases} 120 \pm 10^\circ\text{C} \\ r = 0.018 \end{cases}$ $\begin{cases} 138 \pm 10^\circ\text{C} \\ r = 0.010 \end{cases}$	1 0.3	Mass-spectrometric measurements of Cs 135/Cs 137 abundance ratio in fission-products formed at high flux. Compared with the zero flux value for this ratio, PETRUSKA et al. (1955). Neutron spectra characterised by BIGHAN (1958), but thermal components are probably not truly Maxwellian. Results recalculated with revised data for Co flux monitor
For comparison:				
BERNSTEIN et al. (1956)	(2.70 ± 0.10) (3.03 ± 0.10)	$\begin{cases} g^* = 3/8 \\ g^* = 5/8 \end{cases}$	$T 130^\circ\text{C}$ $r 0$	Evaluated from the resonance parameters for purely Maxwellian spectra; WESTCOTT (1958). Slightly lower values of δ_A would result if the neutron spectrum has a small slowing-down component
SMITH et al. (1959)	(2.99 ± 0.07) (3.38 ± 0.07)	$\begin{cases} g^* = 3/8 \\ g^* = 5/8 \end{cases}$		

6.6 Recommendations for further work

Assessments of the relative importance on the various fission-product poisons in thermal reactors have been given by PATTENDEN & ROSE (1958), PATTENDEN (1959), SHAFTMAN (1959), and WALKER (1960). Four groups of poisons may be distinguished.

- (1) Stable and long-lived fission products with high absorption cross-sections, $\sigma_A \gtrsim 10^4$ barns, namely:



in order of importance.

In a power reactor these fission-products tend fairly quickly to their equilibrium concentrations. Thereafter as may be seen from equation (6.2), their poisoning effect depends only on their fission yields and not on their absorption cross-sections.

- (2) Xe^{135} . This is the most important of the poisons in spite of its short half-life of 9.18 hours. The uncertainties of the cross-section data have been discussed in section 6.5.
- (3) Stable and long-lived fission products with small cross-sections, $\sigma_A \lesssim 1000$ barns. These nuclides approach their saturation concentrations rather slowly so that their poisoning effect depends on their absorption cross-sections as well as their fission-yields. From the papers cited above it appears that the major uncertainties to the poisoning of U233, U235 or Pu239 fuel arise from the following nuclides which are listed roughly in order of importance:

Nuclide	Parameters contributing the largest uncertainties to the poisoning
Pm^{147}	σ_o and Σ'_r
Xe^{131}	σ_o and Σ'_r
Nd^{143}	σ_o and fission-yield, (Σ'_r is small)
Sm^{152}	Σ'_r
Rh^{103}	Σ'_r and fission-yield
Cs^{133}	Σ'_r
Te^{99}	Σ'_r
Nd^{145}	Σ'_r
Ag^{109}	Σ'_r
Ru^{101}	Σ'_r
Zr^{93}	σ_o
Kr^{88}	σ_o
Kr^{104}	σ_o
Pd^{105}	σ_o
Sm^{147}	σ_o
Sm^{148}	σ_o
Sm^{150}	σ_o
Sm^{134}	σ_o
Xe^{85}	σ_o
Kr^{85}	σ_o

- (4) Long-lived and stable fission-products whose absorption cross-sections are unknown. Following is a list of all such fission-products and capture products whose half-lives exceed 1 day. It does not seem likely that there is another significant poison, other than Xe^{135} , with a half-life shorter than 1 day. However the possibility cannot altogether be excluded.

Nuclide	Half-life	Nuclide	Half-life
As^{76}	1.10 day	Sn^{125}	9.5 day
As^{77}	1.62 day	Sn^{126}	2×10^5 yr
Se^{79}	6×10^4 yr	Sb^{122}	2.8
Br^{82}	1.5 day	Sb^{124}	61 day
Rb^{86}	18.7 day	Sb^{125}	2.0 yr
Sr^{89m}	~ 10 day	Sb^{126}	28 day
Nb^{93m}	12 yr	Sb^{127}	3.7 day
Nb^{95m}	3.75 day	Te^{126m}	58 day
Nb^{95}	35 day	Te^{127m}	58 day
Zr^{95}	63 day	Te^{129m}	110 day
Mo^{99}	2.8 day	Te^{131m}	33 day
Ru^{103}	40 day	Te	1.25 day
Ru^{106}	1.0 yr	Te^{132}	3.23 day
Rh^{105}	1.52 day	Xe^{129m}	8.0 day
Pd^{107}	7.5×10^6 yr	Xe^{131m}	12.0 day
Ag^{110m}	270 day	Xe^{133m}	2.2 day
Ag^{111}	7.5 day	Cs^{136}	13 day
Ag^{110}	stable	Ba^{135m}	1.21 day
Cd	stable	Ce^{141}	32 day
Cd^{111}	stable	Ce^{144}	285 day
Cd^{112}	stable	Nd^{147}	11.1 day
Cd^{113m}	5.1 yr	Pm^{148}	5.3 day
Cd^{115m}	43 day	Pm^{149}	44 day
Cd^{116}	stable	Pm^{149}	2.2 day
In^{114}	49 day	Pm^{151}	1.15 day
Sn^{117m}	14 day	Sm^{153}	1.96 day
Sn^{119m}	245 day	Eu^{158}	15 day
Sn^{121m}	>400 day	Tb^{161}	6.9 day
Sn^{121}	1.15 day		
Sn^{123m}	130 day		

In addition to the deficiencies which have been outlined above, perusal of the table of cross-sections will reveal many discrepancies and uncertainties. Most of the early activation data are highly unreliable because the decay schemes of the product nuclei were not well characterised at the time. Many of the resonance integrals have been estimated from very sparse information. The following may be noted particularly:

- (i) Br⁷⁹. Directly measured resonance integral is much smaller than that calculated from the resonance parameters. The activation measurement is complicated by the excitation of isomeric states.
- (ii) Rb. The calculated resonance integrals for Rb⁸⁵ and Rb⁸⁷ are much too small to account for the value ~11 barns obtained by KLIMENTOV & GRYAZEV (1957) for natural Rb.
- (iii) Mo¹⁰⁰. The values of σ_0 and $\Sigma' r$ estimated by WALKER (1960) from the resonance parameters are twice as large as the values measured directly by CABELL (1960).
- (iv) Tc⁹⁹. The data by TATTERSALL et al. (1960) for σ_0 and $\Sigma' r$ appear to be too small.
- (v) In¹¹³. The activation resonance integral is much greater than the calculated from the resonance parameters.
- (vi) Sb. Activation data for Sb in reactor spectra appear too low in comparison with the calculated resonance integrals and thermal cross-sections.
- (vii) Te. The estimates from resonance parameters appear to be inconsistent with the activation and oscillator data.
- (viii) Cs¹³³, La¹³⁹. The activation data in reactor spectra appear too low in comparison with other data, when the strong resonance absorption is taken into account.
- (ix) Ce¹⁴⁰, Ce¹⁴². For each, σ_A and σ_{act} differ by a factor of two.
- (x) Pr¹⁴¹. The directly measured resonance integral, HARRIS et al. (1950a), is only about 1/3rd of the value calculated from the resonance parameters.
- (xi) Pm¹⁴⁷. The available data are discordant and unreliable.
- (xii) Nd¹⁴⁶. The pile oscillator value for σ_0 is much higher than the activation data.
- (xiii) Sm¹⁵². The activation data for δ are erratic.
- (xiv) Eu¹⁵¹. The activation data are very poor.
- (xv) Eu¹⁵², Eu¹⁵⁴, Eu¹⁵⁵. The data are not very reliable.

Reference

- AITKEN K.L., LITTLER D.J., LOCKETT E.E. & PALMER G.H. (1957) J.Nucl.En. 4, 33.
ANDERS E. (1958) Phys. Rev. 110, 427 (1958).
ANDERSON H.L., FERMI E., WATTENBERG A., WEIL G.L. & ZINN W.H. (1944), Phys. Rev. 72, 16 (1947).
ANIKINA M.P., ARON P.M., GORSHKOV V.K., IVANOV R.N., KRIZHARSKY L.M.,
KUKAVADZE G.M., MURIN A.N. REFORMATSKY I.A. & ERSHLER B.V. (1958)
Proc. Geneva II, 15, 446. See also GORSHKOV et al. (1957)
ARNOLD J.R. & SUGARMAN N. (1947) J. Chem. Phys. 15, 703.
BAERG A.P., BROWN F., & LOUNSBURY M. (1958) Can. J. Phys. 36, 863.
BAYLY J.G., BROWN F., HALL G.R. & WALTER A.J. (1958) J. Inorg. Nucl. Chem. 5, 259
BELL P.R., CASSIDY J.M., DAVIS R.C. & KELLEY G.G. (1950) ORNL 940, 30.
See also CASSIDY (1951).
BENOIST P., KOWARSKI L. & NETTER F. (1951) J. Phys. Rad. 12, 584.
BERNSTEIN S., SHAPIRO M.M., STANFORD C.P., STEPHENSON T.E., DIAL J.B., FREED S.,
PARKER G.W., BROSI A.R., HEBERT G.M. & DEWITT T.W. (1956) Phys. Rev. 102, 823.
BERNSTEIN S. & SMITH E.C. (1956) Proc. Geneva I, 4, 153.
BIDINOSTI D.R., FICKEL H.R. & TOMLINSON R.H. (1958) Proc. Geneva II, 15, 459.
BLOCK R.L., SLAUGHTER G.G. & HARVEY J.A. (1959) ORNL 2718, 26.
BOTHE W. (1946) Zeitschr. Naturforsch. 1, 179.
BOWE J.C. & AXEL P. (1951) Phys. Rev. 84, 939.
BUTLEMENT F.D.S. (1949) Phys. Rev. 75, 1276
BUTLEMENT F.D.S. (1950) Proc. Phys. Soc. 63A, 532.
CABELL M.J. (1960) Reactor Sci. 12, 172.
CASSIDY J.M. (1951) Phys. Rev. 83, 483.
COCKING S.J. (1958) NRDC 107, unpublished.
COCKING S.J. (1958) TNCC(UK) 34.
COLMER F.C.W. & LITTLER D.J. (1950) Proc. Phys. Soc. A 63, 1175.
CORNISH F. & AITKEN K.L. (1958), private communication.
Der MATEOSIAN E., GOLDHABER M., MUEHLHAUSE C.O. & McKEOWN M. (1947) Phys. Rev. 72,
1271.
Der MATEOSIAN E. & GOLDHABER M. (1957) Phys. Rev. 108, 766.
DEUTSCH R.W. (1956) Phys. Rev. 104, 555.
EASTWOOD T.A., BAERG A.P., BIGHAM C.B., BROWN F., CABELL M.J. GRUMMITT W.E.,
ROY J.C., ROY L.P. & SCHUMAN R.P. (1958) Proc. Geneva II, 16, 54.
EGELSTAFF P. (1953) AERE N/R 1147.
FICKEL H.R. & TOMLINSON R.H. (1959) Can. J. Phys. 37, 531.
FLAMMERSFELD A. (1952) Zeitschr. Naturforsch. 7a, 295.
FREEDMAN M.S., TURKEVICH A., ADAMS R.M. SUGARMAN N., RAYNOR S. & STANG L.B. (1945)
J. Inorg. Nucl. Chem. 2, 271 (1956).
GLENDENIN L.E. & MARINSKY J.A. (1949) NP 1242.
GORSHKOV V.K., IVANOV R.N., KUKAVADZE G.M. & REFORMATSKY I.A. (1957)
Atomnaya Energiya 3, 11; Soviet J. Atom. En. 3, 729.
GRIMELAND B. (1952) Phys. Rev. 86, 937.
HARRIS S.P. MUEHLHAUSE C.O. & THOMAS G.E. (1950a) Phys. Rev. 79, 11.
HARRIS S., MUEHLHAUSE C.O., RASMUSSEN S., SCHROEDER H.P. & THOMAS G.E. (1950b)
Phys. Rev. 80, 342.
HARVEY J.A., BLOCK R.C., SLAUGHTER G.G., MARTIN W.H. & PARKER G.W. (1958)
Proc. Geneva II, 16, 150.
HAY H.J. (1958) J. Nucl. En. 7, 199.
HAY H.J. & PATTENDEN N.J. (1959) NRDC 113.
HAYDEN J.R., REYNOLDS J.H. & INGHAM M.G. (1949) Phys. Rev. 75, 1500.
HESS D.C. & INGHAM M.G. (1949) Phys. Rev. 76, 300.

- HILL R.D. (1951) Phys. Rev. 83, 865.
- HILL R.D. (1949) Phys. Rev. 76, 333.
- HCAGLAND E.J. & SUGARMAN N. (1945) National Nuclear Energy Series, div.IV, 2, 642, (McGraw-Hill, 1951).
- HOPKINS G.R. & JAMIESON C.P. (1957) J. Appl. Phys. 28, 1362.
- HUGHES D.J., SPATZ W.D.B. & GOLDSTEIN N. (1949) Phys. Rev. 75, 1781.
- HUGHES D.J., GARTH R.C. & LEVIN J.S. (1953) Phys. Rev. 91, 1423.
- HUGHES D.J. & SHERMAN D. (1950) Phys. Rev. 78, 632.
- INGHRAM M.G., HAYDEN R.J. & HESS D.C. Phys. Rev. 79, 271 (1950)
- INGHRAM M.G., HAYDEN R.J. & HESS D.C. (1947) Phys. Rev. 71, 643.
- IVANOV R.N., GORSHKOV V.K., ANIKINA M.P., KUKAVADZE G.M. & ERSHLER B.V. (1957) Atomnaya Energiya 3, 546.
- JONES W.B. (1947) Phys. Rev. 72, 362.
- JOWITT D., PATTENDEN S.K., ROSE H., SMALL V.G. & TATTERSALL R.B. (1958) AERE R/R 2516.
- KATCOFF S. (1945) National Nuclear Energy Series, div.IV, 2, 1400, (McGraw-Hill, 1951).
- KATCOFF S., LEARY J.A., WALSH K.A., ELMER R.A., GOLDSMITH S.S., HALL L.D., NEWBURY E.G., POVELITES J.J. & WADDELL J.S. (1949) J. Chem. Phys. 17, 421.
- KATCOFF S. & WILLIAMS D.C. (1958) J. Inorg. Nucl. Chem. 1, 194.
- KATCOFF S. (1958) Nucleonics 16:4, 78.
- KENNEDY T.J. & THODE H.G. (1958) J. Inorg. Nucl. Chem. 5, 253.
- KERN B.D. & SNELL A.H. (1943) CP 1087, 10. Quoted in NBS Circular 499.
- KLIMENTOV V.B. & GRYAZEV V.M. (1957). Atomnaya Energiya 3, 507; J. Nucl. En. 2, 20.
- LEDDICOTTE G.W. & REYNOLDS S.A. (1949) ORNL 286.
- LEDDICOTTE G.W. (1949) Unpublished quoted in N.B.S. Circular 499
- LYON W.S. & ELRIDGE J.S. (1957) Phys. Rev. 107, 1056.
- MACNAMARA J. & THODE H.G. (1950) Phys. Rev. 80, 296.
- MAKAROV E.F. & SAMOLOVA Z.D. (1958) Fizika i Teplotekhnika Reaktorov, Moscow. Quoted in Nucl. Sci. Abstr. 12, 15821.
- MARINSKY J.A. & GLENDEEN L.E. (1945) National Nuclear Energy Series, div.IV, 2, 1264, (McGraw-Hill, 1951).
- MCREYNOLDS A.W. & ANDERSEN E. (1954) Phys. Rev. 93, 195.
- MEIJER R.R. (1949) Phys. Rev. 75, 773.
- MEINKE W.W. (1953) Phys. Rev. 90, 410.
- MEISTER H. (1958) Zeitschr. Naturforsch. 13a, 820.
- MELAIKA E.A., PARKER M.J., PETRUSKA J.A. & TOMLINSON R.H., (1955) Can. J. Chem. 33, 830.
- MURIN A.N. NEFEDOV V.D., POPOV D.K. & BARANOVSKII V.I. (1957). Atomnaya Energiya 2, 553; J. Nucl. En. 1, 265.
- NELSON C.M., KETELLE B.R. & BOYD G.E. (1950) ORNL 828.
- NEPHEW E.A. (1960) ORNL-2869.
- OVERMAN R.T., FRY L.M., JONES J.W., BALDWIN W.D., LAMB E. & SAVOLAINEN J.E. (1948) ORNL 4.
- PARKER G.W., LANTZ P.M., INGHRAIM M.G., HESS D.C. & HAYDEN R.J. (1947) Phys. Rev. 72, 85.
- PATTENDEN N.J. (1958) Proc. Geneva II, 16, 44.
- PATTENDEN N.J. & ROSE H. (1958) AERE NP/R 2651.
- PATTENDEN N.J. (1959) NRDC 110.
- PATTENDEN N.J. (1960) ORNL 60-8-44.

- PETRUSKA J.A., MELAIKA E.A., & TOMLINSON R.H. (1955) Can. J. Phys. 33, 640.
- POMERANCE H. (1951) Phys. Rev. 83, 641.
- POMERANCE H. (1952) Phys. Rev. 88, 412.
- POMERANCE H. (1955a) ORNL 1879, 50; and (1955b) ORNL 1975, 31.
- PURKAYASTHA B.C. & MARTIN G.R. (1956) Can. J. Chem. 34, 293.
- REYNOLDS J.H. (1950a) ORNL 867, 24; and (1950b) Phys. Rev. 79, 789.
- ROSE H., COOPER W.A. & TATTERSALL R.B. (1958) Proc. Geneva II, 16, 34.
- ROY L.P. & YAFFE L. (1956a) Can. J. Chem. 34, 238. See also EASTWOOD et al. (1958).
- ROY L.P. YAFFE L. (1956b) Can. J. Chem. 34, 1023.
- ROY L.P. & ROY J.C. (1957) Can. J. Phys. 35, 1215. See also EASTWOOD et al. (1958).
- ROY J.C., BERRY P.S. & ROY L.P. (1958) Can. J. Chem. 36, 731. See also EASTWOOD et al. (1958).
- ROY J.C. & WUSCHKE D. (1958) Can. J. Chem. 36, 1424.
- ROY J.C. & ROY L.P. (1959) Can. J. Phys. 37, 907.
- RUSSELL B., SACHS D., WATTENBERG A. & FIELDS R. (1948) Phys. Rev. 73, 545.
- SAILOR V.L. (1953) Phys. Rev. 91, 53.
- SEHGAL M.L., HANS H.S. & GILL P.S. (1959) Nucl. Phys. 12, 261.
- SEREN L., FRIEDLANDER H.N. & TURKEL S.H. (1947) Phys. Rev. 72, 888.
- SHAFTMAN D.H. (1959), Unpublished.
- SHARMA B.L. & POOL M.L. (1959) Bull. Am. Phys. Soc. II: 4, 151. Also SHARMA B.L. (1960) Nuovo Cim. 17, 687.
- SMITH E.C., PAULICKI G.S., THURLOW P.E.F., PARKER G.W., MARTIN W.J., CREEK G.E., LANTZ P.M. & BERNSTEIN S. (1959) Phys. Rev. 115, 1693.
- SMITH R.R. & REEDER S.D. (1955) J. Chem. Phys. 23, 2108.
- SMITH R.R., REEDER S.D. & LEWIS R.H. (1956) J. Chem. Phys. 25, 502.
- STUPEGIA D.C. (1960) Reactor Sci. 12, 16.
- STURM W.J. (1947) Phys. Rev. 71, 757.
- SUGARMAN N. & TURKEVICH A. (1944) National Nuclear Energy Series, div. IV, 9, 1396. (McGraw-Hill, 1951).
- SUGARMAN N. (1949) Phys. Rev. 75, 1473.
- SUNYAR A.W. & GOLDHABER M. (1949) Phys. Rev. 76, 189.
- TATTERSALL R.B., ROSE H., PATTENDEN S.K., & JOWITT D. (1960) Reactor Sci. 12, 32.
- THULIN S. (1955) Ank. Fys. 9, 137.
- TURKEVICH A., ADAM R., FREEDMAN M. & STANG L. (1944) CC 2485, 5. Quoted in NBS Circular 499.
- TURKEVICH A., STEINBERG E.P., FINKLE B. & SUGARMAN N. (1945) National Nuclear Energy Series, div. IV, 9, 1070, (McGraw-Hill, 1951).
- WAHL A.C. (1959) J. Inorg. Nucl. Chem. 10, 1.
- WALKER W.H. (1956) Doctoral thesis at McMaster University, unpublished. Cited by WALKER (1960).
- WALKER W.H. (1960) AECL 1054.
- WALKER W.H. & THODE H.G. (1953) Phys. Rev. 90, 447.
- WESTCOTT C.H. (1958) AECL-670, and (1960) AECL 1101
- WILES D.M. & TOMLINSON R.H. (1955) Can. J. Phys. 33, 133.
- WU C.S., RAINWATER L.J. & HAVENS W.W. (1947) Phys. Rev. 71, 174.
- YAFFE L., SARGENT B.W., KIRSCH M., STANDIL S. & GRUNLUND J.M. (1949) Phys. Rev. 76, 617.